

(Part 2 of 3)

# **Radiological Contamination Control Training for Laboratory Research**

## **Instructor's Guide**



**Office of Environment, Safety & Health  
U.S. Department of Energy  
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DEPARTMENT OF ENERGY - COURSE/LESSON PLAN

**Standardized Core Course Materials**

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<b>Course Goal:</b>	Upon completion of this training, the participant will have a basic understanding of radiological contamination control measures for laboratory research facilities.
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<b>Target Audience:</b>	Individuals who have assigned duties as laboratory researchers.
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<b>Course Description:</b>	This course illustrates and reinforces the skills and knowledge needed to assist personnel with radiological controls for laboratory research facilities.
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<b>Prerequisites:</b>	None
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<b>Length:</b>	2-8 hours (depending on facility-specific information and incorporation of practical exercises)
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Course Objectives Next

**Terminal Objective:** At the end of this course, the participant will understand the basic radiological contamination control measures for working in a research laboratory.

**Enabling Objectives:**

- EO1 DISTINGUISH between ionizing radiation and radioactive contamination.
- EO2 DEFINE
  - Fixed.
  - Removable.
  - Airborne contamination.
- EO3 IDENTIFY the units used to measure radioactive contamination.
- EO4 IDENTIFY causes of radioactive contamination.
- EO5 IDENTIFY methods used to control radioactive contamination.
- EO6 DEFINE Contamination Area, High Contamination Area, and Airborne Radioactivity Area.
- EO7 IDENTIFY the requirements for entry, working in, and exiting Contamination Areas and Airborne Radioactivity Areas.
- EO8 IDENTIFY the proper use of protective clothing.
- EO9 STATE the appropriate response to a spill of radioactive material.

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## DOE-HDBK-1106-97

### Enabling Objectives: (continued)

- E010 IDENTIFY methods for reducing radioactive waste.
  - E011 IDENTIFY the purpose and use of personnel contamination monitors.
  - E012 IDENTIFY the normal methods used for decontamination.
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### Student Materials:

Student's Guide  
Plus any handouts or other materials for facility-specific information or activities.

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Bibliography Next

**Bibliography:**

**DOE standards, handbooks, and technical standards lists (TSLs). The following DOE standards, handbooks, and TSLs form a part of this document to the extent specified herein.**

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DOE (1993), U.S. Department of Energy, "Radiation Protection of the Public and the Environment," DOE Order 5400.5, Ch. 2.

DOE (1998), U.S. Department of Energy, 10 CFR Part 835, "Occupational Radiation Protection."

DOE/CH-9401 (1993), "Performance of Surveys for Unrestricted Release Facility Guidance," R&D Laboratory Working Group (RADWG), Health Physics Procedures (HP) Committee.

DOE (2004a), U.S. Department of Energy, "Radiological Control Standard," (RCS) DOE-STD-1098-1999, Ch. 1.

DOE (2004b), "Environment, Safety, and Health Reporting Requirements," DOE Order 231.1A, Ch. 1.

DOE (1995), EH-412 memorandum (R. Pelletier, Director Office of Environmental Policy and Assistance) to the field, "Application of DOE Order 5400.5 Requirements for Release and Control of Property Containing Radioactive Material."

DOE (1998), U.S. Department of Energy, DOE-HDBK-1130-98, Reaffirmation 2004, "Radiological Worker Training," and DOE-HDBK-1131-98, Reaffirmation 2004, "General Employee Radiological Training."

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**Other government documents, drawings, and publications.** The following government documents, drawings, and publications form a part of this document to the extent specified herein. Unless otherwise indicated, the issues of these documents are those cited in the contracting document.

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NRC (1982), U.S. Nuclear Regulatory Commission, Regulatory Guide 8.23, "Radiation Safety Surveys at Medical Institutions."

NRC (1991), U.S. Nuclear Regulatory Commission, 10 CFR Part 20, "Standards for Protection Against Radiation."

Federal Guidance Report No. 11 (1988), Limiting Values of Radionuclide Intake and Air Concentration and Dose Conversion Factors for Inhalation, Submersion, and Ingestion (based on the 1987 Federal Radiation Protection Guidance), Oak Ridge National Laboratory, 1988.

**Non-Government documents**

LA-4400 (1970), LA-4400, Los Alamos Handbook of Radiation Monitoring.

ANSI 13.12 (1999), "Surface Radioactivity Guides for Materials, Equipment, and Facilities to be Released for Uncontrolled Use," HPS Standards Committee.

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National Council on Radiation Protection and Measurement (NCRP), NCRP No. 106 (1989), "Radiation Protection for Medical and Allied Health Personnel."

Lorenzen (1994), W.A., Ring J.P., "The Management and Operation of a Large Scale Decay-In-Storage Program," paper presented at 27th Mid-Year Topical Meeting of the Health Physics Society, Albany, NY.

Stein (1992), F., "Instructor Competencies: The Standards," International Board of Standards for Training, Performance and Instruction.

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Lesson Summary Next

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## LESSON SUMMARY

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### Introduction

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Welcome students to the course.

Introduce self to the participants and establish rapport.

Define logistics:

- Safety briefing - exits.
  - Restrooms.
  - Hours.
  - Breaks.
  - Sign-in sheets.
  - Test - accountability.
  - End-of-course evaluation.
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### Terminal Objective

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At the end of this course, the participant should be able to understand the basic radiological contamination control measures for working in a research laboratory.

State Enabling Objectives.

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### Course Content

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Briefly review the content of the course, noting that there is a logical sequence (“flow”) to the material covered. As you present the material, relate it to the circumstances they can expect to find in the facility workplace and procedures. (You will be inserting facility-specific laboratory researcher information.)

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Lesson Plan and Instructor’s Notes Next

Lesson Plan	Instructor's Notes
<p><b>I. RADIOLOGICAL CONTAMINATION</b></p> <p>A. Comparison of Radiation and Radioactive Contamination</p> <p>Radiation is energy, contamination is a material. Exposure to radiation does NOT result in contamination.</p> <p>B. Types of Contamination</p> <p>Contamination can be grouped into 3 types:</p> <ul style="list-style-type: none"> <li>• Fixed.</li> <li>• Removable/transferable.</li> <li>• Airborne.</li> </ul> <p>1. Fixed contamination</p> <p>Fixed contamination is contamination that cannot be readily removed from surfaces.</p> <ul style="list-style-type: none"> <li>• It cannot be removed by casual contact, wiping, brushing, or washing.</li> <li>• It may be released when the surface is disturbed (buffing, grinding, using volatile liquids for cleaning, construction, etc.).</li> <li>• Over time it may “weep,” leach, or otherwise become loose or transferable.</li> </ul>	<p>EO1: DISTINGUISH between ionizing radiation and radioactive contamination.</p> <p>EO2: DEFINE fixed, removable, and airborne contamination.</p>

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## 2. Removable/transferable contamination

Removable/transferable contamination is contamination that can be readily removed from surfaces.

- It may be removed or transferred by casual contact, wiping, brushing, or washing.
- Air movement across contamination could cause the contamination to become airborne

## 3. Airborne contamination

Airborne contamination is contamination suspended in air.

This creates a particular hazard because of the possibility of intake by inhalation. Inhalation is the most common mode of uptake of radioactive material in the working environment. In addition to the hazard to the worker, radioactive materials may be carried into ventilation systems, material may be deposited on surfaces over a large area, and there is the potential for releases outside of the facility.

Lesson Plan	Instructor's Notes
<p data-bbox="256 226 760 262">C. Measuring Radioactive Contamination</p> <p data-bbox="306 296 992 394">Because radioactive contamination is radioactive material, the units are the same, e.g., disintegrations per minute (dpm).</p> <p data-bbox="306 428 992 562">When measuring the amount of radioactive contamination (material) on a surface, the units most commonly used are disintegrations per minute per 100 centimeters squared (dpm/100 cm<sup>2</sup>).</p> <p data-bbox="306 596 505 632">1. Direct reading</p> <p data-bbox="331 665 1011 764">Contamination monitors measure radiation emitted by the radioactive material. The units are normally counts (as seen by the monitor) per minute (cpm).</p> <p data-bbox="306 798 1027 863">2. Discuss counts per minute (cpm) versus disintegrations per minute (dpm).</p> <p data-bbox="331 896 1011 1031">There is a direct relationship between the counts recorded and the actual activity (disintegrations) present. The counter efficiency (expressed as the ratio of cpm/dpm) is divided into the measured cpm to obtain the activity.</p>	<p data-bbox="1053 296 1377 394">EO3: IDENTIFY the units used to measure Radioactive contamination.</p> <p data-bbox="1053 896 1377 932">Explain counter efficiency.</p>

Lesson Plan	Instructor's Notes
<p data-bbox="256 264 756 296">D. Causes of Radioactive Contamination</p> <p data-bbox="306 331 990 428">Radioactive material can be spread to unwanted locations. The following are some causes of radioactive contamination:</p> <ol data-bbox="306 464 1015 1268" style="list-style-type: none"> <li data-bbox="306 464 993 531">1. Sloppy work practices, such as cross-contamination of tools, equipment, or workers.</li> <li data-bbox="306 562 961 594">2. Not wearing gloves, or removing them prematurely.</li> <li data-bbox="306 625 846 657">3. Poor housekeeping in contaminated areas.</li> <li data-bbox="306 688 987 756">4. Opening radioactive materials/systems without proper controls.</li> <li data-bbox="306 787 990 854">5. Leaking containers or tears in radiological containers such as barrels, plastic bags, boxes, or protective gear.</li> <li data-bbox="306 886 829 917">6. Spills, glass breakage, and animal fluids.</li> <li data-bbox="306 949 907 980">7. Airborne contamination depositing on surfaces.</li> <li data-bbox="306 1012 1015 1129">8. Not adhering to standard laboratory procedures (such as not checking gloves after handling radioactive materials or working in a potentially contaminated area).</li> <li data-bbox="306 1161 630 1268">9. Emergencies including: <ul data-bbox="355 1203 591 1268" style="list-style-type: none"> <li data-bbox="355 1203 456 1234">• Fire.</li> <li data-bbox="355 1234 591 1268">• Earthquake, etc.</li> </ul> </li> </ol>	<p data-bbox="1052 264 1380 331">EO4: IDENTIFY causes of radioactive contamination.</p> <p data-bbox="1052 562 1406 659">Optional method may include listing trainees' response on board or flip chart.</p>

Lesson Plan	Instructor's Notes
<p data-bbox="256 264 818 294">D. Indicators of Possible Area Contamination</p> <p data-bbox="332 331 924 394">The following are some indicators of possible area contamination:</p> <ol data-bbox="332 432 1024 1115" style="list-style-type: none"> <li data-bbox="332 432 683 462">1. Visual indicators, such as: <ul data-bbox="381 468 821 531" style="list-style-type: none"> <li data-bbox="381 468 781 497">• Leaks, spills, standing liquids.</li> <li data-bbox="381 501 821 531">• Damaged radiological containers.</li> </ul> </li> <li data-bbox="332 569 943 632">2. Detection of contamination or elevated radiation levels including: <ul data-bbox="381 638 1024 1115" style="list-style-type: none"> <li data-bbox="381 638 1016 667">• Spurious or unexplained personnel contamination.</li> <li data-bbox="381 705 938 768">• Radioactivity observed in bioassay samples collected.</li> <li data-bbox="381 806 959 869">• Higher than normal background on personnel contamination survey devices.</li> <li data-bbox="381 907 1024 970">• Higher than normal background radiation levels on area monitors and air samples.</li> <li data-bbox="381 1008 902 1115">• Routine radiation contamination surveys conducted by the Radiological Control Organization.</li> </ul> </li> </ol>	<p data-bbox="1055 699 1341 930">Presence of higher than background levels of radioactivity in bioassay samples indicates radiological control measures are not being effectively implemented</p>



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F. Primary Reasons for Contamination Control

## 1. Protection of the worker

Measures to control radioactive contamination are implemented to protect workers by:

- Minimizing the chance of inhalation or ingestion of radioactive/hazardous material.
- Eliminating or reducing external radiation dose rates.
- Reducing worker discomfort by minimizing the use of personal protective clothing and/or respirators.

## 2. Radioactive materials may enter the body by:

Inhalation (the most common pathway).

- Cuts/wounds (i.e., sharp instrument punctures).
- Absorption (skin, mucous membranes, eyes).
- Ingestion (biting nails, applying cosmetics, eating, drinking, or smoking either in the lab or outside without monitoring or washing hands).

Lesson Plan	Instructor's Notes
<p data-bbox="306 264 699 296">3. Protection of the environment</p> <p data-bbox="355 331 943 394">Measures to control radioactive contamination are implemented to protect the environment by:</p> <ul data-bbox="355 401 932 569" style="list-style-type: none"><li data-bbox="355 401 932 464">• Controlling the release of radioactivity in the environment.</li><li data-bbox="355 499 932 569">• Minimizing the amount of radioactive waste generated.</li></ul> <p data-bbox="306 604 813 636">4. Protection of the facility and programs</p> <p data-bbox="355 672 943 735">Measures to control radioactive contamination are implemented to protect facilities and programs by:</p> <ul data-bbox="355 741 956 978" style="list-style-type: none"><li data-bbox="355 741 956 804">• Eliminating or minimizing the spread of contamination.</li><li data-bbox="355 840 956 903">• Preventing cross-contamination and the loss of experimental results.</li><li data-bbox="355 938 797 978">• Meeting regulatory requirements.</li></ul> <p data-bbox="306 1050 1005 1182">Also, note that resources applied to cleanup reduce the resources available for other program goals. Additionally, serious contamination events may detract from maintaining good public relations.</p>	

## Lesson Plan

## Instructor's Notes

## G. Radiological Contamination Control Measures

Radiological contamination control measures should address:

1. Characteristics of radionuclides used:
  - Type of radiation emitted
  - Energy of radiation emitted
  - Half-life
2. Preparation of areas and materials - including:
  - Marking, labeling, and posting of areas and materials.
  - Personnel protective equipment type, availability, and use.
  - Storage and containment of radioactive/hazardous materials.
3. Good work practices - including:
  - Special precautions for handling liquids.
  - Special precautions for handling sharps.
  - Clean up the work area at the end of the job or end of the day, whichever is first.
4. Radioactive waste management.
5. Radiation monitoring (including interpretation of meter readings) during and at completion of work. If a problem is detected or suspected, notify the Radiological Control Organization.

Lesson Plan	Instructor's Notes
<p data-bbox="355 262 594 296">6. Decontamination.</p> <p data-bbox="355 329 683 363">7. Regulatory requirements.</p> <p data-bbox="355 396 654 430">8. Training requirements.</p> <p data-bbox="355 464 1029 531">The following sections discuss in detail characteristics of radionuclides and contamination control measures.</p>	

Lesson Plan	Instructor's Notes
<b>II. CHARACTERISTICS OF COMMONLY USED RADIONUCLIDES</b>	Select those radionuclides from Appendix A that are applicable to your facility and add applicable radionuclide information.

Lesson Plan	Instructor's Notes
<p><b>III. PREPARATION OF WORK AREA AND MATERIALS</b></p> <p>E. Appropriate Selection of Work Area</p> <p>The work station should not present an exposure potential to another individual within the laboratory or to the adjacent laboratory. The work station should not conflict with other work within the laboratory (i.e., strong gamma emitters near low background counting equipment, etc.).</p> <p>F. Preparation of Work Areas</p> <ol style="list-style-type: none"> <li>1. Minimize area <p>Confine operations involving radioactive materials to a small as space as practicable.</p> </li> <li>2. Clear area <p>Clear area of extraneous items and material.</p> </li> <li>3. Work surface <p>Cover area as appropriate. Diaper paper should be placed absorbent side up.</p> </li> <li>4. Containment <p>Use trays when appropriate.</p> </li> </ol>	<p>Material in this Section may be deleted if it is covered in other training.</p> <p>EO5: IDENTIFY methods used to control Radioactive contamination.</p>

Lesson Plan	Instructor's Notes
<p>5. Waste</p> <p>Receptacles for radioactive waste should be located by the work station so that you may conveniently dispose of waste without further contamination of the work area.</p> <p>C. Preparation of Equipment/Materials</p> <p>1. Assemble survey meters</p> <p>The survey meter should be turned on and located in close proximity to the work station. A preoperational check is necessary before use to ensure the meter is working properly. Position the detector so it is directed toward your work area. This will enable you to conveniently monitor your hands as you work and also can indicate when materials are removed from shielded containers. Always work with the audio turned on. Your safety and the safety of others must take priority over the concept that the sound may be disturbing to others. Know the location of the nearest phones. Post the Radiological Control Organization phone number nearby.</p> <p>2. Equipment preparation</p> <p>Use dedicated equipment/tools when appropriate. Cover/tape tools or equipment used during the job to minimize radioactive contamination.</p>	<p>State that later in the course the selection and use of the proper instruments will be covered (Section VI).</p>

Lesson Plan	Instructor's Notes
<p>3. Assemble materials and supplies</p> <p>Those supplies that would minimize small spills of radioactive materials should be within arm's reach while handling unsealed radioactive materials.</p> <p>4. Marking and labeling</p> <ul style="list-style-type: none"> <li>• The area/material is marked and labeled as appropriate (typically by Radiological Control personnel).</li> <li>• Do <b>NOT</b> discard intact radioactive labels/markings in normal trash! Remove or deface labels before discarding boxes, etc.</li> </ul> <p>5. Assemble spill materials (as appropriate)</p>	<p>Show appropriate labels, tape, etc.</p> <p>Discuss consequences if radiation label is found in municipal waste stream.</p>
<p>D. Shielding</p> <p>1. Placement</p> <p>Placement of shielding materials is critical to both your safety and that of your colleagues. Work stations that require the use of shielding should be located where there would be no worker on the opposite side of the workbench, such as in corners and against walls.</p>	



Lesson Plan	Instructor's Notes
<p>2. Penetration through walls</p> <p>Be certain to consider what (or who) is on the other side of the wall. However, if this can not be accommodated, shielding should be considered for the work station on the opposite side of the workbench where your colleague may be working.</p> <p>3. Beta and Gamma Emitters</p> <p>When shielding for both beta and gamma emitters, the shielding for the beta emitters should be first. The beta shield (plastic, wood, foil) should be closer to the radiation source to minimize the production of X-rays from the beta emitter interacting with the lead (gamma) shield.</p> <p>4. Considerations</p> <p>Some considerations for use of shielding were addressed above. Because issues involving shielding can be complex, always consult the Radiological Control Organization before using shielding.</p> <p>E. Ventilation Control</p> <p>1. Airflow</p> <p>Airflow should be <u>from</u> the areas of least contamination <u>to</u> areas of most contamination (e.g., clean to contaminated to highly contaminated areas) such as fume hoods, gloveboxes, etc.</p>	

Lesson Plan	Instructor's Notes
<p>2. Pressure differential</p> <p>Slight negative pressure is maintained in buildings/rooms where potential contamination exists.</p> <p>3. High Efficiency Particulate Air (HEPA) filtration</p> <p>HEPA filters, which remove radioactive particles from the air, may be used. This is commonly required for higher levels of airborne radioactivity such as concentrations exceeding 10 percent of the Derived Air Concentration (DAC) for a particular radionuclide. Charcoal filters are required for specific radionuclides such as iodine. Ventilation system requirements should be determined in consultation with the Radiological Control Organization.</p> <p>A DAC is the radionuclide airborne concentration. Breathing an air concentration of 1 DAC for 1 working year (2,000 hours) will result in committed dose equivalent equal to an annual limit, i.e., 5 rem whole-body or 50 rem to any organ or tissue.</p> <p>4. Flow rate</p> <p>Always check the flow rate or pressure in ventilated enclosures before starting operations. Air flow is easily measured with an inexpensive velometer.</p>	<p>Insert facility-specific requirements.</p>

Lesson Plan	Instructor's Notes
<p data-bbox="253 262 651 296">F. Posting of Radiological Areas</p> <p data-bbox="310 331 976 464">10 CFR 835.602 requires that each access point to a controlled area be appropriately posted. Further, 835.602 requires that each access point to a radiological area be appropriately posted.</p> <p data-bbox="310 499 959 596">Posting of radiological areas is typically a Radiological Control Organization function. Workers should become aware of the posting requirements.</p> <p data-bbox="310 632 927 695">DOE has designated the following areas as requiring posting:</p> <ol data-bbox="310 699 662 732" style="list-style-type: none"> <li data-bbox="310 699 662 732">1. Radioactive Material Area</li> </ol> <p data-bbox="355 768 1016 865">An area or structure where radioactive material, exceeding the values provided in 10 CFR 835 Appendix E, is used, handled, or stored.</p> <p data-bbox="355 934 711 968">The posting/sign will indicate:</p> <p data-bbox="355 1003 919 1037"><b>“CAUTION, RADIOACTIVE MATERIAL”</b></p> <p data-bbox="355 1073 967 1169">Additional posting is not required if the Radioactive Material Area is inside a Contamination, High Contamination, or Airborne Radioactivity Area.</p>	<p data-bbox="1057 262 1409 359">Materials in this Section need not be repeated if it is covered in other training.</p> <p data-bbox="1057 632 1398 833">Reference 10 CFR 835 and exemptions from posting. Reference Glossary for definition of radioactive material, controlled area, and radiological area.</p> <p data-bbox="1057 903 1409 1199">Discuss that entry requirements may be included on radiological postings, (e.g., TLD required, monitor upon exiting and/or protective clothing required). Insert facility-specific posting requirements and examples here.</p>

Lesson Plan	Instructor's Notes
<p>2. Contamination Area</p> <p>Any area where removable contamination levels are greater than the values specified in Appendix D of 10 CFR 835, but less than or equal to 100 times those levels.</p> <p>The posting/signs will indicate:</p> <p><b>“CAUTION, CONTAMINATION AREA”</b></p>	<p>EO6: DEFINE Contamination Area, High Contamination Area, and Airborne Radioactivity Area. Reference: 10 CFR 835.2(a), 835.603(e), and Appendix D of 10 CFR 835.</p>

<b>Abbreviated Table of Contamination Values</b> (See 10 CFR 835 Appendix D for complete listing)		
NUCLIDE	REMOVABLE	TOTAL dpm/ 100 cm <sup>2</sup>
I-125, I-129	20	500
I-131, I-133	200	1,000
Beta/gamma	1,000	5,000
Tritium	10,000	N/A

Lesson Plan	Instructor's Notes
<p>3. High Contamination Area</p> <p>Any area where removable contamination levels are greater than 100 times the values listed in Appendix D of 10 CFR 835.</p> <p>The posting/sign will indicate:</p> <p><b>“DANGER, HIGH CONTAMINATION AREA</b> Radiological Work Permit (RWP) Required for Entry.”</p>	<p>Reference: 10 CFR 835.2(a)</p>
<p>4. Airborne Radioactivity Area</p> <p>Any area where the measured concentration of airborne radioactivity, above natural background, exceeds or is likely to exceed the derived air concentration (DAC) values listed in Appendix A or Appendix C of 10 CFR 835 or where an individual in the area without respiratory protection could receive an intake exceeding 12 DAC-hours in a week.</p> <p>The posting/sign will indicate:</p> <p><b>“CAUTION, AIRBORNE RADIOACTIVITY AREA”</b></p> <p>A DAC is the radionuclide airborne concentration. Breathing an air concentration of 1 DAC for 1 working year (2,000 hours) will result in committed dose equivalent equal to an annual limit, i.e., 5 rem whole body or 50 rem to any organ or tissue.</p>	<p>Reference: 10 CFR 835.2(a) and 835.603(d)</p>

Lesson Plan	Instructor's Notes
<p data-bbox="256 262 935 296">G. Labeling of Radioactive Materials and Other Postings</p> <p data-bbox="310 329 911 363">1. <b>“CAUTION, RADIOACTIVE MATERIAL”</b></p> <p data-bbox="355 396 967 527">Equipment, components, and other items that are radioactive, potentially radioactive, or have been in contact with radioactive contamination or activation sources.</p> <p data-bbox="310 562 992 661">2. <b>“CAUTION, INTERNAL CONTAMINATION” or “CAUTION, POTENTIAL INTERNAL CONTAMINATION”</b></p> <p data-bbox="355 697 1003 764">Equipment, components, and other items with actual or potential internal contamination.</p> <p data-bbox="310 800 894 833">3. <b>“CAUTION, FIXED CONTAMINATION”</b></p> <p data-bbox="355 867 946 934">Components, equipment, or other items with fixed contamination.</p> <p data-bbox="310 970 764 1003">4. Facility-Specific Postings/Labeling</p> <p data-bbox="355 1037 889 1066">These may include Radiological Buffer areas.</p>	<p data-bbox="1055 262 1333 296">Reference: 10 CFR 835</p> <p data-bbox="1055 562 1328 630">Reference: RCS, Ch. 1, Article 412</p> <p data-bbox="1055 800 1328 867">Reference: RCS, Ch. 1, Article 412</p> <p data-bbox="1055 970 1403 1268">Add facility-specific postings/labeling. Discuss differences between posting and labeling. Areas that can be physically entered are posted. Equipment, components, and radioactive sources are labeled. Reference: RCS Article 233</p>

Lesson Plan	Instructor's Notes
<p><b>IV. CONDUCT OF WORK - GOOD PRACTICES</b></p> <p>A. Personal Preparation</p> <p>Ensure that you are ready to work and that you have the following:</p> <ul style="list-style-type: none"> <li>• Training to meet entry requirements.</li> <li>• Work permits, procedures, etc.</li> <li>• Dosimetry.</li> <li>• Personal protective equipment.</li> </ul> <p>B. Requirements of Posted Contamination Areas</p> <p>1. Requirements for entry into posted contamination areas.</p> <p>The RadCon Standard recommends individuals allowed unescorted entry into Contamination Areas be provided the following:</p> <ul style="list-style-type: none"> <li>• Radiological Worker II training.</li> <li>• Worker's signature on the Radiation Work Permit, as applicable.</li> <li>• Protective clothing/equipment as required by the Radiation Work Permit.</li> <li>• Personnel dosimetry, as appropriate.</li> <li>• Pre-job briefing for High Contamination and Airborne Radioactivity Areas.</li> </ul>	<p>Reference: RCS, Rev 1, Article 335</p> <p>EO7: IDENTIFY the requirements for entry, working in, and exiting Contamination Areas and Airborne Radioactivity Areas.</p>

Lesson Plan	Instructor's Notes
<p>2. Requirements for working in posted contamination areas</p> <p>The RadCon Standard recommends that individuals allowed unescorted entry into Contamination Areas:</p> <ul style="list-style-type: none"> <li>• Avoid unnecessary contact with contaminated surfaces.</li> <li>• When possible wrap or sleeve materials and/or equipment brought into the area.</li> <li>• Do not touch unexposed skin surfaces. This could result in skin contamination.</li> </ul> <p>Smoking, eating, chewing, drinking, and putting on makeup could result in ingesting radioactive material; for this reason, these activities are not allowed in Contamination Areas.</p> <p>3. Requirements for exiting posted contamination areas</p> <p>The RadCon Standard recommends that individuals allowed unescorted entry into Contamination Areas:</p> <p>a. Exit only at step-off pad.</p> <ul style="list-style-type: none"> <li>• A step-off pad provides a “barrier” between contaminated and other areas to prevent or control the spread of contamination between areas. Correct use of step-off pads is included in the practical factors exercise.</li> </ul>	<p>Reference: RCS, Ch. 1, Article 335</p>



Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>• If more than one step-off pad is used, the final step-off pad is “clean,” outside the exit point, and adjacent to the boundary of the Contamination Area.</li> </ul> <p>b. Remove protective clothing carefully and slowly.</p> <p>Loose contamination on the clothing can be dislodged causing a possible spread of contamination or even potential inhalation if contamination becomes airborne.</p> <p>c. Perform a personal survey. If contamination is indicated:</p> <ul style="list-style-type: none"> <li>• Stay in the area.</li> <li>• Notify Radiological Control personnel.</li> <li>• Take action to minimize cross-contamination (e.g., put a glove on a contaminated hand or tape over contamination on clothing).</li> </ul> <p>d. Tools or equipment being removed from a posted area must be monitored prior to release.</p> <p>e. After exiting and monitoring yourself, it is a good practice to wash your hands.</p>	<p>In some cases, Radiological Control personnel may perform the survey. Tritium cannot be detected with standard survey meters.</p> <p>Reference: 10 CFR 835.1101 requirements.</p> <p>Insert facility-specific procedures.</p>

Lesson Plan	Instructor's Notes
<p>C. Dosimetry</p> <p>Always have proper personnel monitoring that might include:</p> <ol style="list-style-type: none"> <li>1. Whole body <p>Whole-body dosimeter such as a thermoluminescent dosimeter (TLD) or film badge.</p> </li> <li>2. Extremity monitoring <p>Finger rings, if handling high contact dose rate materials such as P-32.</p> </li> </ol> <p>D. Personnel Protective Clothing (Anti-C)</p> <p>The degree of clothing required is dependent on the work area, radiological conditions, and the nature of the job. The use of personnel protective clothing and equipment is the least desired option. Use of engineering controls such as gloveboxes or fumehoods is preferred. Standard clothing requirements for biomedical research laboratory work include:</p> <ul style="list-style-type: none"> <li>• Lab coats with long sleeves that are buttoned or otherwise closed.</li> </ul>	<p>Insert facility-specific dosimetry requirements</p> <p>EO8: IDENTIFY the proper use of protective clothing.</p>

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>• Surgeon's gloves; the gloves may be needed to protect against radioactive contamination, as well as other lab hazards applicable, such as acids and caustics.</li> <li>• Closed-toed shoes.</li> <li>• Safety glasses or equivalent for eye protection from eye hazards including radiological hazards, such as from P-32.</li> </ul> <p>1. Proper use of protective clothing</p> <ul style="list-style-type: none"> <li>• Inspect all protective clothing for rips, tears, holes, or wear prior to use.</li> <li>• Personal effects such as watches, rings, jewelry, etc. should not be worn.</li> <li>• After donning protective clothing, such as anti-contamination clothing, proceed directly from the dress-out area to the work area. In general, a lab coat is sufficient to protect the individual at most biomedical research laboratories.</li> <li>• Avoid getting lab coats wet. Wet lab coats provide a means for contamination to reach the skin/clothing.</li> <li>• Contact Radiological Control personnel if clothing becomes ripped, torn, etc. during operations.</li> </ul>	<p>DISCUSS the use of gloves.</p> <p>DEMONSTRATE how to change gloves without contaminating fingers or wrists.</p> <p>DISCUSS how and why contamination can reach the skin and clothing.</p>

Lesson Plan	Instructor's Notes
<p>2. Eye protection</p> <p>Safety glasses, goggles, or face shields must be worn to prevent eye contamination in the event of splashes or droplet contamination. In addition, eye protection will provide protection from moderate to high energy beta radiation, such as betas emitted from P-32.</p> <p>3. Respiratory equipment</p> <p>Respiratory equipment is used to prevent the inhalation of radioactive materials. This training course does not qualify a worker to wear respiratory equipment. Ventilation design should eliminate the need to use respiratory equipment except in extreme cases.</p> <p>E. Storage and Containment of Radioactive Material</p> <p>Containment generally means using vessels, trays, diaper paper, bench tops, etc. to contain contamination.</p>	<p>EO5: IDENTIFY methods used to control Radioactive contamination.</p>

Lesson Plan	Instructor's Notes
<ol style="list-style-type: none"><li>1. Storage areas<ol style="list-style-type: none"><li>a. Store large bottles and containers close to the floor.</li><li>b. Shelves should:<ul style="list-style-type: none"><li>• Be secured (bolted) to a wall.</li><li>• Have lips or restraining cords to prevent bottles from falling.</li></ul></li><li>c. Storage area should be well lit, properly ventilated, and have an even temperature.</li></ol></li><li>2. Radioactive materials should be properly stored:<ul style="list-style-type: none"><li>• In unbreakable containers; if not possible, in secondary containment (the secondary containment should be able to contain the entire volume of the primary container).</li><li>• In stable containers with secure means of closing.</li><li>• Away from sinks and drains or other possible pathways that do not collect in retention tanks.</li><li>• Protected from adverse environmental factors.</li><li>• Away from combustibles and other fire sources.</li></ul></li></ol>	

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>• Protected from “unauthorized relocation,” this may include locked refrigerators and storage cabinets.</li> <li>• With the outside of container clearly labeled with contents.</li> <li>• With provided instructions to open containers.</li> </ul> <p>3. Posting and labeling of storage areas</p> <p>Room access and cabinets, refrigerators, freezers, etc. that house the container should be posted or labeled “Caution Radioactive Material” or “Caution Radioactive Material Storage Area.”</p> <p>4. Chemical considerations for storage</p> <p>Segregate incompatibles and store by hazard class. Appendices B and C address chemical storage in more detail.</p> <p>F. Good Housekeeping</p> <p>“Good housekeeping” is the prime factor in an effective contamination control program and it involves the interactions of all groups within the facility. Each individual must be dedicated to keeping “his house clean” to help control the spread of contamination.</p>	<p>Storage of chemicals may be addressed here if not covered in other training.</p> <p>EO5: IDENTIFY methods used to control Radioactive contamination.</p>

Lesson Plan	Instructor's Notes
<p>G. RadCon Required Actions and Good Practices</p> <ol style="list-style-type: none"> <li>1. BELIEVE! labels and posted areas.</li> <li>2. Avoid Contamination and Airborne Radioactivity Areas. These areas should be isolated from routine operations.</li> <li>3. Treat Contamination Areas as if everything was contaminated.</li> <li>4. Minimize the number of items carried or placed into potentially contaminated areas.</li> <li>5. Use proper and functional radiation detection instrumentation.</li> <li>6. Do not eat, drink, apply makeup, etc.</li> <li>7. Always wash hands upon completion of work.</li> </ol> <p>H. Special Precautions for Liquids</p> <p>Radioactive solutions are a potential source of radioactive contamination if they are spilled or allowed to evaporate. A particular concern of a spill is that it may be a source of airborne radioactivity. In addition, when radioactive material is in a solution, it can be carried to places not normally accessible, e.g., under equipment.</p>	<p>It is important to remove labels and postings as soon as the radioactive materials are no longer present so that employees will believe that the labels are correct at all times.</p> <p>Remind attendees that there are release requirements.</p>

Lesson Plan	Instructor's Notes
<p>1. Handling liquids</p> <p>Standard good practices for handling liquids include:</p> <ul style="list-style-type: none"> <li>• Use appropriate gloves for liquids being handled.</li> <li>• Protect personal clothing.</li> <li>• Work in tray with absorbent paper.</li> <li>• Use mechanical pipettes and dilutors (<b>NEVER</b> pipette by mouth).</li> <li>• Work in a properly vented area.</li> <li>• Report any spills or suspected spills</li> </ul> <p>2. Preventing spills</p> <p>The best way to handle a spill is to prevent it in the first place by:</p> <ul style="list-style-type: none"> <li>• Storing materials unless in use.</li> <li>• Limiting quantities to what is needed.</li> <li>• Keeping work area clean and free of obstructions.</li> <li>• Using stable containers with secure means of closure.</li> </ul>	<p>Note: If the liquids could generate airborne radioactivity, additional posting and monitoring may be required.</p> <p>Discuss facility-specific reporting requirements.</p>



Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>• Avoiding unstable (top heavy) containers or arrangements.</li> <li>• Using secondary containment for liquids.</li> </ul> <p>3. Leaking containers</p> <ul style="list-style-type: none"> <li>• Report all suspected leaks immediately to the Radiological Control Organization.</li> <li>• If the material is highly toxic, evacuate everyone from the area.</li> <li>• Leaking containers should be placed in a fume hood if it can be done safely</li> </ul> <p>4. Handling spills</p> <p>One simple method utilized for response to spills is the acronym SWIMS, which stands for:</p> <ul style="list-style-type: none"> <li>• Stop the spill.</li> <li>• Warn others.</li> <li>• Isolate the area.</li> <li>• Minimize exposure.</li> <li>• Secure the ventilation system. If the spill involves volatile chemical or volatile or gaseous radionuclides, the ventilation may need to be left on. Add facility-specific information.</li> </ul> <p>As previously discussed, report the spill to the Radiological Control Organization.</p>	<p>EO9: STATE the appropriate response to a spill of radioactive material.</p> <p>Discuss and demonstrate facility/room-specific response to a spill.</p>

Lesson Plan	Instructor's Notes
<p><b>V. RADIOACTIVE WASTE MANAGEMENT</b></p> <p>A. Segregation</p> <p>Segregate waste by waste stream category and half-life to facilitate storage, minimization, and disposal.</p> <p>B. Waste Storage</p> <p>Each laboratory should have a designated location for storing waste. Radioactive waste should be stored separately from hazardous waste. This location should be out of the way of normal lab activities, but easily accessible, recognizable, and properly labeled and shielded.</p> <p>Liquid waste materials should be kept in secondary containers and segregated by hazard class. Secondary containers may be lab trays or any device that will contain 110 percent of the largest container.</p> <p>C. Sharps</p> <p>Contaminated syringes, glass pipettes, and other sharp items must be placed in a specifically designed, rigid container.</p>	<p>EO10: IDENTIFY methods for reducing radioactive waste.</p> <p>Waste Stream Segregation includes maintaining separation between hazardous and radioactive wastes.</p>

Lesson Plan	Instructor's Notes
<p>D. Methods for Minimizing Waste</p> <ol style="list-style-type: none"> <li>1. Minimize waste generation               <ol style="list-style-type: none"> <li>a. <u>Confine operations</u>: Confine operations with radioactive materials to as small an area as possible.</li> <li>b. <u>Minimize materials</u>: Minimize materials introduced into radioactive material handling areas.</li> <li>c. <u>Segregate</u>: Segregate clean materials from radioactive materials. Do not dispose of clean materials in radioactive waste containers.</li> <li>d. <u>Good housekeeping</u>: Contamination control measures, such as covering benches, etc., generate waste. On the other hand, decontamination generates a great deal of waste. Good housekeeping, following procedures, minimizing bench areas and secondary containment can reduce the amount of coverings required.</li> </ol> </li> </ol>	<p>Ask attendees for reasons and methods.</p>

Lesson Plan	Instructor's Notes
<p>2. Storage for Decay</p> <p>a. <u>Storage</u>: Some radionuclides have a short half-life and can be stored, with appropriate DOE approval, for decay. Normal storage times are 10 half-lives. The waste must be surveyed prior to disposal to ensure it is below disposal criteria.</p> <p>b. <u>Substitution</u>: Substitute shorter-lived for longer-lived radionuclides, if possible.</p>	<p>Reference: Lorenzen (1994)</p>
<p>3. Disposal via Sanitary Sewer</p> <p>Disposal of small quantities via sanitary sewer is available to some facilities. Add facility specific information.</p>	<p>Reference: DOE (1993)</p> <p>DOE allows discharge via sanitary sewer per Order 5400.5, Ch. 2, Chapter II, 3.d. Include facility-specific procedures.</p>
<p>4. Disposal of specific waste per 10 CFR 20.2005</p> <p>In accordance with Part 20.2005, NRC licensees may dispose of the following as if it were not radioactive:</p>	<p>NRC licensees and most agreement States allow for limited quantities of radionuclides to be disposed via the sanitary sewer per 10 CFR 20.2003 or the agreement State equivalent regulation. Reference: NRC (1991).</p>

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>a. <u>Liquid scintillation counting (LSC) media</u>: Liquid scintillation counting media containing 0.05 microcuries or less per gram of medium of H-3 or C-14.</li> <li>b. <u>Animal carcasses</u>: Animal carcasses containing 0.05 microcuries or less per gram of animal tissue (averaged over the weight of the entire animal) of H-3 or C-14.</li> </ul> <p>5. Volume reduction</p> <ul style="list-style-type: none"> <li>a. <u>Compaction</u>: May produce reduction factors of up to 5 to 1.</li> <li>b. <u>Shredding</u>: May produce reduction factors of up to 12 to 1.</li> <li>c. <u>Incineration</u>: Difficult under present regulations and political climate; allowed under 10 CFR 20.2004. Currently, scintillation fluids are incinerated by commercially licensed vendors.</li> </ul>	<p>Discuss concerns of LSC sewer disposal (clog drains, exposure to LSC fluid). Need to follow facility-specific procedures. Encourage use of non-tolulene-based LSC.</p> <p>Animal carcasses can't be used for human food or animal feed.</p> <p>Reference: Woehr (1994)</p>

Lesson Plan	Instructor's Notes
<p data-bbox="261 264 464 289">E. Mixed Waste</p> <p data-bbox="310 331 1008 527">Mixed waste is exceedingly difficult, if not impossible, to dispose of at this time. Currently, DOE has a self-imposed moratorium on the off-site shipment of RCRA/TSCA waste suspected of having radioactivity. This moratorium was instituted by the DOE Office of Waste Management (EM-30).</p> <ol data-bbox="310 569 1024 1272" style="list-style-type: none"> <li data-bbox="310 569 1024 730">1. The EM-30 Performance Objective (PO) for Certification of Non-radioactive Hazardous Wastes was developed in 1991 and is currently under revision. This PO was developed to guide DOE sites in addressing the issue of hazardous waste that contains added radioactivity.</li> <li data-bbox="310 772 1024 1272">2. Ways to avoid generating mixed waste: <ul data-bbox="358 804 1024 1272" style="list-style-type: none"> <li data-bbox="358 804 1024 867">• Use non-hazardous cleaning materials for decontamination whenever possible.</li> <li data-bbox="358 909 1024 972">• Segregate “radioactive only” from “hazardous only” at the source.</li> <li data-bbox="358 1014 1024 1108">• Explore the use of other materials that are non-hazardous for use in radiological areas to prevent the generation of mixed waste.</li> <li data-bbox="358 1150 1024 1272">• Discontinue use of non-biodegradable (organic solvent based) liquid scintillation media. Biodegradable liquid scintillation media are available.</li> </ul> </li> </ol>	<p data-bbox="1057 264 1365 289">Reference: Stevens (1994)</p>

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"><li>• Some States are more restrictive than the U.S. Environmental Protection Agency (EPA) in their listing of those scintillation cocktails that are biodegradable.</li><li>• Organic solvents with a flash point below 60°C (140°F) may be classified as “ignitable,” thus creating a mixed waste where disposal may not be possible. If the flash point is above 140°F, the organic solvent may not be considered as ignitable; however, these materials must be handled and disposed of with extreme caution.</li></ul> <p>3. Other methods for facilitating disposal:</p> <ul style="list-style-type: none"><li>• Do not combine solvents with metals; disposal is very difficult. Examples are lead or mercury combined with solvents.</li><li>• Generally, it is a good idea to separate organics and inorganics whenever possible to facilitate disposal.</li></ul>	





Lesson Plan	Instructor's Notes
<p>Most hand-held survey instruments are calibrated to a Cesium-137 source. Correction factors for the specific radionuclide being monitored should be known when surveying. Often the survey instrument will over or under respond when monitoring for beta or alpha radiation.</p> <p>3. Pre-operational Checks</p> <p>Perform pre-operational checks before work:</p> <ul style="list-style-type: none"> <li>• Confirm calibration is current.</li> <li>• Verify that battery is OK.</li> <li>• Perform an audio check (audio response is immediate, while needle response takes time to stabilize.</li> <li>• Ensure instrument responds to source.</li> <li>• Verify that background count rate is normal.</li> </ul>	<p>DEMONSTRATE preoperational checks.</p> <p>Explain that there is a statistical variation in count rates and demonstrate this range daily.</p>

Lesson Plan	Instructor's Notes
<p>B. Conducting Surveys - General</p> <ol style="list-style-type: none"> <li>1. Survey hands before picking the probe up.</li> <li>2. Hold the probe approximately ½ inch from surface being surveyed.</li> <li>3. Move probe <u>slowly</u> over surface to be surveyed, approximately 2 inches per second.</li> </ol>	<p>Demonstrate general surveys.</p>
<p>C. Area Surveys</p> <ol style="list-style-type: none"> <li>1. Frequently monitor work areas.</li> <li>2. Monitor upon completion of work (or prior to taking a break and leaving the work area).</li> <li>3. Monitor at least every 2 hours for work in progress.</li> <li>4. Wipe surveys should be performed on equipment and areas where survey instruments are not adequate to monitor contamination.</li> </ol>	<p>Explain facility policy. Clarify who is responsible to perform these surveys.</p> <p>Explain wipe survey.</p>
<p>D. Personnel Surveys</p> <ol style="list-style-type: none"> <li>1. Proceed to survey in the following typical order: <ul style="list-style-type: none"> <li>• Head (pause at mouth and nose for approximately 5 seconds).</li> <li>• Neck and shoulders.</li> </ul> </li> </ol>	<p>Demonstrate personnel surveys.</p>

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"><li>• Arms (pause at each elbow), hands, wrists; especially where gloves end.</li><li>• Chest and abdomen.</li><li>• Back, hips, and seat of pants.</li><li>• Legs and cuffs.</li><li>• Shoe tops.</li><li>• Shoe bottoms (pause at sole and heel).</li></ul> <p>2. The whole-body survey should take approximately 3 minutes. A full whole-body survey or frisk is not generally necessary for routine bench-top operations unless a spill occurs or contamination is found on the hands or face. The survey should be done before removing the lab coat and repeated on personal clothing if contamination is found.</p>	

Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"><li>3. If the count rate increases during frisking (such as the audible signal), pause for 5-10 seconds over the area to provide adequate time for instrument response.</li><li>4. Carefully return the probe to holder.</li><li>5. Keep the instrument close to the work area to facilitate frequent checking of hands and fingers.</li></ul> <p>E. Detection of Contamination</p> <p>If contamination is indicated:</p> <ul style="list-style-type: none"><li>1. Remain in the immediate area.</li><li>2. Notify Radiological Control personnel.</li><li>3. Minimize cross-contamination (such as putting a glove on a contaminated hand until decontamination can be attempted).</li></ul>	

Lesson Plan	Instructor's Notes
<p>F. Release of Materials</p> <p>1. Release to Controlled Areas</p> <p>Release from potentially contaminated areas to Controlled Areas is covered by 10 CFR 835.</p> <p>Equipment and materials released from a potentially contaminated area must be surveyed and released under a formal program.</p> <p>If surveys indicate the presence of removable contamination at levels greater than 10 CFR 835 Appendix D, the materials may be moved to another posted area. Appropriate controls must be established and should include:</p> <ul style="list-style-type: none"> <li>• Surveys of materials before movement.</li> <li>• Containment of materials during transit.</li> <li>• Establishment of approved transit routes.</li> <li>• Survey of transit route after movement (if materials exhibited removable contamination).</li> </ul>	<p>Insert facility-specific information as appropriate (if laboratory personnel do not conduct release surveys, it may be appropriate to just cover their responsibilities).</p> <p>Reference: DOE/CH-9401 (1993)</p> <p>Reference: 10 CFR 835.1101(b)</p>

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Lesson Plan	Instructor's Notes
<p>If surveys indicate presence of fixed contamination only (removable contamination less than 10 CFR 835 Appendix D values), the items may be released to the controlled area if:</p> <ul style="list-style-type: none"><li>• Routine surveys are performed to ensure contamination remains fixed to surface.</li><li>• The item is clearly labeled or tagged to warn others of the contamination.</li><li>• Written procedures are established to control such items.</li></ul> <p>2. Unrestricted Release</p> <p>Unrestricted release is addressed in DOE Order 5400.5, Ch. 2. See Appendix F for more information.</p> <p>3. Techniques</p> <p>Monitoring techniques for release of materials is covered in Appendix D.</p>	<p>Reference: 10 CFR 835.1101(c)</p> <p>Additional guidance was provided to the field by EH-412 on Nov. 17, 1995.</p>

Lesson Plan	Instructor's Notes
<p><b>VII. DECONTAMINATION</b></p> <p>Decontamination is the removal of radioactive materials from locations where it is not wanted. This does not result in the disappearance of radioactive material, but involves the removal of the radioactive materials to another location.</p> <p>A. Decontamination or Not</p> <p>If the presence of loose contamination is discovered, decontamination is a valuable means of control.</p> <p>In some situations, this is not always possible.</p> <ol style="list-style-type: none"> <li>1. Economical conditions <p>Cost of time and labor to decontaminate location outweighs the hazards of the contamination present.</p> </li> <li>2. Radiological conditions <p>Radiation dose rates or other radiological conditions present hazards that far exceed the benefits of decontamination.</p> </li> </ol> <p>B. Preventive Methods</p> <ol style="list-style-type: none"> <li>1. Identifying and repairing leaks before they become a serious problem.</li> </ol>	<p>DISCUSS who makes the decision when to decontaminate.</p>

Lesson Plan	Instructor's Notes
<p>2. Changing out gloves or protective gear as necessary to prevent cross-contamination of equipment.</p> <p>C. Skin Contamination</p> <p>Skin contamination normally does not cause physical injury to the skin. Some nuclides and chemical forms allow absorption through the skin (i.e., iodine and tritium). Strong beta emitters may present a hazard to the skin.</p> <p>1. Concerns of skin contamination are:</p> <ul style="list-style-type: none"> <li>• Cross-contamination by touching.</li> <li>• Absorption through the skin.</li> <li>• Threat of uptake by ingestion, touching face, etc.</li> </ul> <p>2. Skin decontamination</p> <p>Intact skin is an excellent barrier, so use gentle methods to decontaminate.</p> <p>Normally, mild soap and lukewarm water are used to decontaminate personnel. Good practices include:</p> <ul style="list-style-type: none"> <li>• Do not abrade skin.</li> </ul>	<p>EO12: IDENTIFY the normal methods used for decontamination</p> <p>Discuss the need to involve RadCon personnel as appropriate. Refer to facility specific requirement.</p>



Lesson Plan	Instructor's Notes
<ul style="list-style-type: none"> <li>• Do not chap skin by cold water or harsh chemicals.</li> <li>• Avoid hot water because it will open pores.</li> </ul> <ol style="list-style-type: none"> <li>3. If skin contamination remains, a common procedure is to wear surgeon's gloves overnight to induce sweat that will lift contamination from the skin. The decision to wear gloves to induce sweat must be made by the responsible Health Physicist. This practice requires detailed documentation and procedural guidance.</li> </ol> <p>D. Material Decontamination</p> <p>Material decontamination is the removal of radioactive materials from tools, equipment, floors, and other surfaces in the work area.</p> <ol style="list-style-type: none"> <li>1. Establish controls to prevent spread of contamination.</li> <li>2. A high priority is to prevent airborne radioactivity.</li> <li>3. Decontaminate from areas of low to high contamination (exception is when potential for airborne is high).</li> <li>4. Decontaminate from top to bottom so that contamination will not run down on the clean surface.</li> <li>5. Only make one pass, then discard or turn wipe to a clean surface (don't recontaminate area).</li> <li>6. The Radiological Control Organization will make the final determination if the material has been adequately decontaminated.</li> </ol>	<p>EO12: IDENTIFY the normal methods used for decontamination.</p>

Lesson Plan	Instructor's Notes
<b>VIII. FACILITY-SPECIFIC REQUIREMENTS</b>  Insert facility-specific information.	

Lesson Plan	Instructor's Notes
<p data-bbox="196 258 812 325"><b>IX. CONTAMINATION CONTROL LESSONS LEARNED</b></p> <p data-bbox="256 359 873 394">Present lessons learned that are relevant to your site.</p>	

Lesson Plan	Instructor's Notes
<p><b>X. SUMMARY - Review of Objectives</b></p> <p>EO1 DISTINGUISH between ionizing radiation and radioactive contamination.</p> <p>EO2 DEFINE</p> <ul style="list-style-type: none"> <li>• Fixed.</li> <li>• Removable.</li> <li>• Airborne contamination.</li> </ul> <p>EO3 IDENTIFY the units used to measure radioactive contamination.</p> <p>EO4 IDENTIFY causes of radioactive contamination.</p> <p>EO5 IDENTIFY methods used to control radioactive contamination.</p> <p>EO6 DEFINE Contamination Area, High Contamination Area, and Airborne Radioactivity Area.</p> <p>EO7 IDENTIFY the requirements for entry, working in, and exiting Contamination Areas and Airborne Radioactivity Areas.</p>	

## DOE-HDBK-1106-97

Lesson Plan		Instructor's Notes
EO8	IDENTIFY the proper use of protective clothing.	
EO9	STATE the appropriate response to a spill of radioactive material.	
EO10	IDENTIFY methods for reducing radioactive waste.	
EO11	IDENTIFY the purpose and use of personnel contamination monitors.	
EO12	IDENTIFY the normal methods used for decontamination.	

## GLOSSARY

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Airborne radioactive material or airborne radioactivity means radioactive material dispersed in the air in the form of dusts, fumes, particulates, mists, vapors, or gases.

Annual Limit On Intake (ALI): The derived limit for the amount of radioactive material taken into the body of an adult worker by inhalation or ingestion in a year. ALI is the smaller value of intake of a given radionuclide in a year by the reference man (ICRP Publication 23) that would result in a committed effective dose equivalent of 5 rems (0.05 sievert) or a committed dose equivalent of 50 rems (0.5 sievert) to any individual organ or tissue. ALI values for intake by ingestion and inhalation of selected radionuclides are based on Table 1 of the U.S. Environmental Protection Agency's Federal Guidance Report No. 11, Limiting Values of Radionuclide Intake and Air Concentration and Dose Conversion Factors for Inhalation, Submersion, and Ingestion, published September 1988. This document is available from the National Technical Information Service, Springfield, VA.

Becquerel (Bq): The SI unit for activity equivalent to 1 nuclear disintegration per second.

Beta Decay: Radioactive decay in which a beta particle is emitted. This transformation changes only the atomic number of the nucleus, raising or lowering the atomic number ( $Z$ ) by one for emission of a negative or positive beta particle, respectively.

Beta Particle: Charged particle emitted from the nucleus during radioactive decay, having a mass and charge equal to that of an electron.

Bioassay: The determination of kinds, quantities, or concentrations, and, in some cases, locations of radioactive material in the human body, whether by direct measurement or by analysis, and evaluation of radioactive materials excreted or removed from the human body.

Biological Half-Life: See Half-Life Biological.

**GLOSSARY (continued)**

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Characteristic X-ray: X-rays that are characteristic of the element in which they are produced. Their emission results from the rearrangement of electrons in the shells of excited atoms.

Contamination: Undesired (e.g., radioactive or hazardous) material that is deposited on the surface of, or internally ingrained into, structures or equipment, or that is mixed with another material.

Radioactive Contamination: A radioactive substance dispersed in materials or places where it is undesirable.

Fixed Contamination: Radioactivity remaining on a surface after repeated decontamination attempts fail to significantly reduce the contamination level.

Removable Contamination: That fraction of the radioactive contamination present on a surface that can be transferred to a swipe tab by rubbing with moderate pressure.

Surface Contamination: The deposition and attachment of radioactive materials to a surface, also the resulting deposit.

Continuous Air Monitor (CAM): Instrument that continuously samples and measures the levels of airborne radioactive materials on a “real time” basis and has alarm capabilities at preset levels.

Controlled Area: Any area to which access is managed to protect individuals from exposure to radiation and/or radioactive materials.

Curie: The unit of activity equal to a rate of  $3.7 \times 10^{10}$  nuclear disintegrations per second.

Decontamination: The reduction or removal of contaminating radioactive material from a structure, area, object, or person. Decontamination may be accomplished by treating the surface to remove or decrease the contamination, or by letting the material stand so that the radioactivity is decreased as a result of natural decay.

Derived Air Concentration (DAC): The airborne concentration that equals the ALI divided by the volume of air breathed by an average worker for a working year of 2,000 hours (assuming a breathing volume of 2400 m<sup>3</sup>). For the radionuclides listed in Appendix C of 10 CFR 835, the air immersion DACs were calculated for a continuous, non-shielded exposure via immersion in a semi-infinite atmospheric cloud. The value is based upon the derived airborne concentration found in Table 1 of the U.S. Environmental Protection Agency's Federal Guidance Report No. 11, Limiting Values of Radionuclide Intake and Air Concentration and Dose Conversion Factors for Inhalation, Submersion, and Ingestion, published September 1988. This document is available from the National Technical Information Service, Springfield, VA.

**GLOSSARY (continued)**

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Disintegration, Nuclear: A spontaneous nuclear transformation (radioactivity) characterized by the emission of energy and/or mass from the nucleus. When numbers of nuclei are involved, the process is characterized by a definite half-life.

Dose Terms:

Committed Dose Equivalent: The calculated dose equivalent projected to be received by a tissue or organ over a 50-year period after an intake of radionuclide into the body. It does not include contributions from external dose.

Committed Effective Dose Equivalent: The sum of the committed dose equivalents to various tissues in the body, each multiplied by its weighting factor.

Cumulative Annual Effective Dose Equivalent: The sum of the annual effective dose equivalents recorded for an individual for each year of employment.

Dose Equivalent: The product of absorbed dose (D) in rad (or gray) in tissue, a quality factor (Q), and other modifying factors (N).

Effective Dose Equivalent: The summation of the products of the dose equivalent received by specified tissues of the body ( $H_T$ ) and the appropriate weighting factor ( $w_T$ )--that is,  $H_E = \sum w_T H_T$ . It includes the dose from radiation sources internal and/or external to the body.

Dosimeter: A portable instrument for measuring and registering the total accumulated dose to ionizing radiation.

Dosimetry: The theory and application of the principles and techniques involved in the measurement and recording of radiation doses. Its practical aspect is concerned with the use of various types of radiation instruments with which measurements are made.



## GLOSSARY (continued)

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Dose Rate: The radiation dose delivered per unit of time. Measured, for example, in rad per hour.

Effective Half-Life: See Half-Life, Effective.

External Radiation: Exposure to ionizing radiation when the radiation source is located outside the body.

Flash Point: The minimum temperature at which a substance gives off flammable vapor that will ignite if in contact with spark or flame.

Fume Hood: Ventilated containment space, enclosed on five sides, with the sixth side covered by a movable glass or plastic window to allow access and to maintain sufficient inflow of air and splash control to protect the worker from the hazardous materials handled inside.

Gamma Ray: Very penetrating electromagnetic radiation of nuclear origin. Except for its origin, it is identical to an X-ray.

Geiger-Mueller Counter: A radiation detection and measuring instrument. It consists of a gas-filled tube containing electrodes, between which there is an electrical voltage but no current flowing. When ionizing radiation passes through the tube, a short, intense pulse of current passes from the negative electrode to the positive electrode and is measured or counted. The number of pulses per second measures the intensity of radiation. It is sometimes called simply a Geiger counter, or a G-M counter.

Gray (Gy): The SI unit for absorbed dose. One gray is equivalent to one Joule per kilogram or 100 rad.

Half-Life Biological ( $T_{\text{bio}}$ ): The time required for the body to eliminate one-half of an administered dose of any substance by regular processes of elimination. This time is usually the same for both stable and radioactive isotopes of a particular element. The biological half-life of tritium is 10 days, whereas the physical half-life is 12.3 years.

**GLOSSARY (continued)**

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Half-Life, Effective ( $T_{\text{eff}}$ ): The time required for the amount of a radioactive nuclide deposited in a living organism to be diminished 50 percent as a result of the combined action of radioactive decay  $T_{1/2}$  and biological elimination  $T_{\text{bio}}$ .

$$T_{\text{eff}} = T_{\text{bio}} \times T_{1/2} / (T_{\text{bio}} + T_{1/2})$$

Half-Life, Physical ( $T_{1/2}$ ): The time in which half the atoms of a particular radioactive substance disintegrate to another nuclear form. Measured half-lives vary from millionths of a second to billions of years.

Half-Value Layer (HVL): Thickness of a specified substance which, when introduced into the path of a given beam of radiation, reduces the exposure rate by one half.

Health Physicist: A person trained to advise on operating procedures for minimizing radiation exposures, perform radiation surveys, oversee radiation monitoring, and estimate the degree of radiation hazard.

Health Physics: The science concerned with recognition, evaluation, and control of health hazards from ionizing radiation.

Health, Radiological: The art and science of protecting human beings from injury by radiation, as well as promoting better health through beneficial applications of radiation.

High Efficiency Particulate Air (HEPA): An air filter generally rated as being capable of removing at least 99.97 percent of the particulate material in an air stream.

Indirect Bioassay: The assessment of radioactive material deposited in the body by detection of radioactivity in material excreted or removed from the body.

Intake: The quantity of material (activity or mass) initially taken into the body. (For example, in the case of inhalation, the intake includes the quantity of material immediately exhaled.)

## GLOSSARY (continued)

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Internal Emitter: A term used for a radionuclide deposited in the body.

In-vitro Methods: Detection of radiations emitted by radioactive materials excreted or removed from the body, using radiochemical and/or radioanalytical techniques.

In-vivo Methods: Detection of radiations emitted by radioactive materials deposited in the body, usually by whole body (or critical organ) counting techniques.

Ionization (Ion) Chamber: An instrument that detects and measures ionizing radiation by measuring the electrical current that flows when radiation ionizes gas in a chamber, making the gas a conductor of the electricity.

Ionizing Radiation: Any electromagnetic or particulate radiation capable of producing ions (either directly or indirectly) in its passage through matter.

Irradiation: Exposure to radiation.

Isotope: One of two or more atoms with the same number of protons, but different numbers of neutrons in their nuclei. Isotopes have very nearly the same chemical properties.

keV: The symbol for one thousand-electron-volts (1,000 eV).

Kilo: Symbol k. A prefix indication base unit is to be multiplied by 1,000.

Kilovolt (kV): A unit of electrical potential equal to 1,000 volts.

Lead Equivalent: The thickness of lead affording the same attenuation, under specified conditions, as the material in question.

License: Written authorization issued to the licensee by the NRC or agreement State to perform specific activities related to the possession and use of byproduct, source, or special nuclear material.

## GLOSSARY (continued)

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MeV: A unit of energy. The energy acquired by an electron accelerated through a potential difference of one million volts.

Micro-: A prefix that divides a basic unit into one million parts. Often used with activity such as microcurie.

Milli-: A prefix that divides a basic unit by 1,000. Often used with dose and activity such as millirem or millicurie.

Million Electron Volts (MeV): Energy equal to that acquired by a particle with one electronic charge in passing through a potential difference of one million volts.

Minimum Detectable Activity (MDA): The lowest amount of any specific radiation that can be detected with a particular level of statistical significance above background levels.

Nano-: A prefix that divides a basic unit by one billion. Often used in measurements of activity such as nanocurie.

Nuclide: A species of atom having a specified number of neutrons and protons in its nucleus.

Personnel Monitoring: Monitoring any part of individuals, their breaths, or excretions, or any part of their clothing to determine the amount of radioactivity present in or on an individual.

Pico-: A prefix that divides a basic unit by one trillion. Often used in measurements of activity such as picocurie.

Proportional Counter: An instrument in which an electronic detection system receives pulses that are proportional to the number of ions formed in a gas-filled tube by ionizing radiation.

Prospective Monitoring: Routine workplace and personnel monitoring for possible intakes or radioactive materials. Prospective monitoring will typically include air monitoring, surface contamination surveys, and bioassay. Any prospective monitoring results above Investigation Levels will trigger retrospective monitoring.

**GLOSSARY (continued)**

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Quality Factor: The principal modifying factor used to calculate the dose equivalent from the absorbed dose; the absorbed dose (expressed in rad or gray) is multiplied by the appropriate quality factor (Q).

Rad: A unit of absorbed dose. The word comes from the acronym Radiation Absorbed Dose and is equivalent to 100 ergs per gram. It does not take into account the biological effect resulting from the absorbed dose.

Radioactive Material: Radioactive material includes any material, equipment, or system component determined to be contaminated or suspected of being contaminated. Radioactive material also includes activated material, sealed and unsealed sources, and material that emits radiation.

Radioactivity: The process whereby certain nuclides undergo spontaneous disintegration in which energy is liberated, generally resulting in the formation of new nuclides. The process is accompanied by the emission of one or more types of radiation, such as alpha particles and gamma photons.

Radiochemical: A molecule or a chemical compound or substance containing one or more radioactive atoms.

Radiological Area: Any area within a controlled area that must be posted as a “radiation area,” “high radiation area,” “very high radiation area,” “contamination area,” “high contamination area,” or “airborne radioactivity area” in accordance with 10 CFR 835.603.

Radiological Buffer Area (RBA): An intermediate area established to prevent the spread of radioactive contamination and to protect personnel from radiation exposure.

Radiological Work Permit (RWP): Permit that identifies radiological conditions, establishes worker protection and monitoring requirements, and contains specific approvals for radiological work activities. The Radiological Work Permit serves as an administrative process for planning and controlling radiological work and informing the worker of the radiological conditions.

**GLOSSARY (continued)**

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Radionuclide: A radioactive (unstable) nuclide.

Radioisotope: An unstable isotope of an element that decays or disintegrates spontaneously, emitting radiation. Approximately 5,000 natural and artificial radioisotopes have been identified.

Reference Man: A hypothetical individual whose characteristics are often used to estimate radiation dose. Reference Man is to be 20-30 years of age, 170 cm (5 ft 10 in) in height, weighing 70 kg (160 lb); and living in a climate with an average temperature of from 10° to 20° C. Reference Man is a Caucasian and is Western European or North American in habitat and custom.

Rem: A unit of dose equivalent. The word comes from the acronym Roentgen Equivalent Man and takes into account the biological effect from an absorbed dose of radiation.

Retrospective Monitoring: Retrospective monitoring is a series of measurements made after an intake is suspected to confirm the intake and assess any doses that may result from the intake.

Roentgen: The unit for exposure. It is that amount of gamma or X-rays required to produce ions carrying 1 electrostatic unit of electrical charge in 1 cubic centimeter ( $2.58 \times 10^{-4}$  coulomb/kg) of dry air under standard conditions.

Scintillation Detector: The combination of phosphor, photomultiplier tube, and associated electronic circuits for counting light emissions produced in the phosphor by ionizing radiation.

Sealed radioactive source means a radioactive source manufactured, obtained, or retained for the purpose of utilizing the emitted radiation. The sealed radioactive source consists of a known or estimated quantity of radioactive material contained within a sealed capsule, sealed between layer(s) of non-radioactive material, or firmly fixed to a non-radioactive surface by electroplating or other means intended to prevent leakage or escape of the radioactive material. Sealed radioactive sources do not include reactor fuel elements, nuclear explosive devices, and radioisotope thermoelectric generators.

## GLOSSARY (continued)

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Shielding: Any material or obstruction that absorbs radiation and thus tends to protect personnel or materials from the effects of ionizing radiation.

SI: The International System of Units (“Le Systeme International d'Unites”) as defined by the General Conference of Weights and Measures in 1960. These units are generally based on the meter/kilogram/second units, with special quantities for radiation including the becquerel, gray, and sievert.

Sievert (Sv): The SI unit of dose equivalent. It is equivalent to 100 rem.

Specific Activity: The total activity of a given nuclide per gram of material. Specific activity is a function of half-life and is therefore also unique to each radionuclide. There are approximately 10,000 curies in a gram of tritium and 1 curie in a gram of radium-226.

Survey Meter: An instrument used to monitor the presence of radioactivity by detecting the radiation emitted during the radioactive decay.

Tenth Value Layer (TVL): Amount of shielding material required to reduce radiation exposure by a factor of 10. One TVL is equal to 3.3 HVL.

Tissue Equivalent Material: Material made up of the same elements in the same proportions as they occur in a particular biological tissue.

Thermoluminescent Dosimeters (TLD): Dosimeters made of certain crystalline materials that are capable of both storing a fraction of absorbed ionizing radiation and releasing this energy in the form of visible photons when heated. The amount of light released can be used as a measure of radiation exposure to these crystals.

Tritium: The hydrogen isotope with one proton and two neutrons in the nucleus. Tritium is radioactive and has a half-life of 12.3 years.

**GLOSSARY (continued)**

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Uptake: Quantity of a radionuclide taken up by the systemic circulation, e.g., by injection into the blood, by absorption from compartments in the respiratory or gastrointestinal tracts, or by absorption through the skin or through wounds in the skin.

Weighting Factor: The fraction of the overall health risk, resulting from uniform, whole body irradiation, attributable to specific tissue (T). The dose equivalent to tissue, T, is multiplied by the appropriate weighting factor to obtain the effective dose equivalent to that tissue.

Whole-Body Counter: A device used to identify and measure the radiation in the body (body burden) of human beings and animals; it uses heavy shielding to keep out background radiation and ultrasensitive scintillation detectors and electronic equipment.

Whole-Body Counting: A technique to determine the internally deposited radionuclides within the body by measuring with an external radiation detector the photons emitted. Results are generally expressed in the form of percent of the ALI for the nuclides in question. This technique can identify and measure accurately normal body radiations as well as those that are taken into the body due to such things as injection, ingestion, and inhalation from atmospheric releases, medical diagnostic and therapeutic techniques, etc.

Whole Body Dose Equivalent: The dose equivalent that results when the whole body is irradiated and taken, when the irradiation is uniform, as equivalent to the effective dose equivalent.

X-rays: Penetrating electromagnetic radiation having wavelengths shorter than those of visible light, usually produced by bombardment of a metallic target with fast electrons in a high vacuum. In nuclear reactions, it is customary to refer to photons originating in the nucleus as gamma rays, and those originating in the extra nuclear part of the atom as X-rays.



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## APPENDIC A

### CHARACTERISTICS OF COMMONLY USED RADIONUCLIDES

#### Instructor's Notes

Select those radionuclides that are applicable to your facility and add applicable radionuclide information.

Effective half-life values are from LA-4400 (see reference LA-4400 1970)

#### TRITIUM (H-3)

Tritium is a low-energy beta emitter and cannot be monitored directly with a thin-window G-M probe. Monitoring is normally performed by taking a swipe of the area and counting the swipe in a liquid scintillation counter.

1. Maximum energy: 0.018 MeV (average energy is 0.006 MeV or about 1/3 the maximum energy).
2. Maximum range in air: 1/4 inch (6 mm).
3. Maximum range in water:  $6 \times 10^{-3}$  mm.
4. International Atomic Energy Agency (IAEA) toxicity classification: Low.

Individuals who handle large quantities of tritium may benefit from the Article 663 course Radiological Training for Tritium Facilities.

**Instructor's Notes**

5. Physical half-life ( $T_{1/2}$ ): 12.35 years.
6. Effective half-life ( $T_{\text{eff}}$ ): 10 days (the time it takes for  $\frac{1}{2}$  of the material to be eliminated from the body by both biological processes and radioactive decay).
7. Critical organ: Whole body (the part of the body where the most limiting dose is delivered).
8. Personnel monitoring: Bioassay - urinalysis, NOT detected with a dosimeter (thermoluminescent dosimeter (TLD) or film).
9. Annual Limit of Intake (ALI): Tritiated water: 80 mCi ( $3 \times 10^9$  Bq) by inhalation or ingestion.
10. Shielding: None (the low-energy beta is not very penetrating).
11. Special Considerations:
  - Cannot be measured directly with a thin window G-M probe (standard survey meter).

Elemental tritium is not taken into the body and is assigned an ALI based on a TEDE resulting from lung exposure. However, in many environments elemental tritium converts rapidly to tritiated water vapor.

**Instructor's Notes**

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- Can be absorbed through the skin.
- Many compounds readily penetrate gloves and skin. Handle these compounds remotely, wear two pairs of gloves, and change the outer layer at least every 20 minutes.
- Tritiated DNA precursors are considered more toxic than tritiated water. However, they are generally less volatile and do not present a significantly greater hazard.

**CARBON-14 (C-14)**

Carbon-14 is a low-energy beta emitter (about 10 times more energetic than tritium). C-14 is not easily detected with a handheld survey instrument such as a thin-window G-M (the efficiency is ~ 10%). Monitoring is normally performed by taking a swipe of the area and counting the swipe in a liquid scintillation counter.

1. Maximum energy: 0.156 MeV (the average energy is 0.052 MeV).
2. Maximum range in air: 9 inches (24 cm).
3. IAEA toxicity classification: Medium-low.
4. Physical half-life: 5,730 years.
5. Effective half-life: ( $T_{\text{eff}}$ ): 12 days.
6. Critical organ: Whole body and the body fat.
7. Personnel monitoring: Bioassay - urinalysis and/or breath measurements ( $\text{CO}_2$ ), NOT detected with a dosimeter (TLD or film).
8. ALI:
  - 2 mCi ( $7 \times 10^7$  Bq) -labeled organic compounds by inhalation or ingestion.
  - 2 Ci ( $7 \times 10^{10}$  Bq)  $\text{CO}$  by inhalation.
  - 200 mCi ( $7 \times 10^9$  Bq)  $\text{CO}_2$  by inhalation.

9. Shielding: 3mm of plexiglass (if needed) – thicker plexiglass may be used for rigidity.

**Instructor's Notes**

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10. Special Considerations:
- Detection of C-14 by radiation survey instruments requires special care due to the low efficiency of detection.
  - Some C-14-labeled compounds may penetrate gloves and skin. Handle these compounds remotely, wear two pairs of gloves and change the outer layer frequently.
  - Special caution should be observed when handling C-14-labeled halogenated acids.

**SODIUM-22 (Na-22)****Instructor's Notes**

Sodium-22 is a positron emitter (positive beta particle/electron) and high-energy gamma emitter. It also emits an annihilation photon when the positive electron is annihilated with a negative electron, producing pure energy.

Sodium-22 is detected with a thin-window G-M probe, sodiumiodide scintillation counter, or liquid scintillation detector.

1. Energy:
  - Maximum beta energy: 0.546 MeV, average energy 0.182 MeV.
  - Gamma energy: 1.275 MeV.
  - Annihilation photon: 0.511 MeV.
2. Maximum beta range in air: 4.7 feet (1.4 m).
3. Unshielded dose rate from 1mCi point source at ½ inch (1cm): 11.8 rad/hr.
4. IAEA toxicity classification: High-medium.
5. Physical half-life: 950 days.
6. Effective half-life ( $T_{\text{eff}}$ ): 10.9 days.
7. Critical organ:
  - Whole body for intake of transportable compounds.

**Instructor's Notes**

- Lungs for inhalation.
  - Lower large intestine for ingestion.
8. Personnel monitoring: dosimeter and finger rings, uptakes may be determined by urinalysis.
9. ALI:
- 0.6 mCi ( $2 \times 10^7$  Bq) by inhalation, clearance in weeks.
  - 0.4 mCi ( $1 \times 10^7$  Bq) by ingestion.
10. Shielding:
- Half-value layer (the thickness required to attenuate the dose rate by  $\frac{1}{2}$ ) is 0.26 inches (6.5mm) of lead.
  - Multi-hundred mCi quantities need to be completely surrounded by beta shielding material to prevent the betas from escaping and creating a source of secondary annihilation radiation outside the shielding.



**Instructor's Notes**

11. Special considerations:

- Near an unshielded Na-22 source, dose rates due to beta radiation can be much higher than dose rates due to gamma radiation.
- Avoid direct eye exposure by interposing transparent shielding or indirect viewing.
- Avoid skin dose by indirect handling.

**PHOSPHORUS-32 (P-32)**

Phosphorus-32 is a high-energy beta emitter that may create a whole body, skin, and an eye hazard. Most common means of detection is with a thin-window probe or liquid scintillation.

**Instructor's Notes**

1. Maximum energy: 1.71 MeV; the average energy is 0.570 MeV.
2. Maximum range in air: 19 feet (6 m).
3. Maximum range in tissue: 8 mm.
4. IAEA toxicity classification: Medium-low.
5. Physical half-life: 14.29 days.
6. Effective half-life ( $T_{\text{eff}}$ ): 10-14 days.
7. Critical organ:
  - Bone for transportable compounds.
  - Lung and lower large intestine are critical organs for inhalation and ingestion, respectively.
8. Personnel monitoring: Dosimeter and finger rings, uptakes may be determined by urinalysis.

**Instructor's Notes**

9. ALI:
- 4mCi ( $1 \times 10^7$  Bq) by inhalation, clearance in weeks.
  - 4mCi ( $1 \times 10^7$  Bq) by ingestion.
10. Shielding:  $\frac{1}{2}$  inch (1.2 cm) of plexiglass (that will shield the beta particles and minimize the production of bremsstrahlung).
- P-32 betas will travel:
- 19 feet in air.
  - 0.8 cm in tissue.
  - 0.7 cm in plexiglass.
  - 0.3 cm in aluminum.
11. Special considerations:
- A high local dose can be received if the radioactive material is touched and allowed to remain in contact with the skin.
  - Do not work over an open container, the eyes can receive a substantial beta dose.
  - Safety glasses can provide eye protection.

- Contamination is easily detected with G-M thin-window probe.
- Bremsstrahlung radiation will be a consideration for millicurie quantities.
- Radwaste containers may need to be shielded with plexiglass.

**Instructor's Notes**

Typical dose rates from 0.1 mCi ( $4 \times 10^6$  Bq):

- 3 mrad/hr at 1 cm.
- 0.03 mrad/hr at 10 cm.
- 0.002 mrad/hr at 40 cm.

**PHOSPHORUS-33 (P-33)**

Phosphorus-33 is a low-energy beta emitter. Most common means of detection is with a thin-window probe or liquid scintillation.

1. Maximum energy: 0.248 MeV; the average energy is 0.083 MeV.
2. Maximum range in air: 1.5 feet (0.5 m).
3. Maximum range in tissue: 1 mm.
4. IAEA toxicity classification: Medium-low.
5. Physical half-life: 24.4 days.
6. Effective half-life ( $T_{\text{eff}}$ ): 10-24 days.
7. Critical organ:
  - Bone for transportable compounds.
  - Lung and lower large intestine are critical organs for inhalation and ingestion, respectively.
8. Personnel monitoring: NOT detected with a dosimeter (TLD or film) dosimeter and finger rings, uptakes may be determined by urinalysis.

**Instructor's Notes**

9. ALI:
- 3 mCi ( $1 \times 10^8$  Bq) by inhalation, clearance in weeks.
  - 6 mCi ( $2 \times 10^8$  Bq) by ingestion.
10. Shielding: 3mm of plexiglass.
11. Special considerations:

Detection of P-33 by radiation survey instruments requires special care due to the low efficiency of detection.

**SULFUR-35 (S-35)**

Sulfur-35 is a low-energy beta emitter similar to carbon-14. Most common means of detection is with liquid scintillation.

1. Maximum energy: 0.167 MeV (the average energy is 0.056 MeV).
2. Maximum range in air: 10 inches (24 cm).
3. Maximum range in tissue: 0.32mm.
4. IAEA toxicity classification: Medium-low.
5. Physical half-life: 87.4 days.
6. Effective half-life ( $T_{\text{eff}}$ ): 77 days.
7. Critical organ: Whole body and testis.
8. Personnel monitoring: Bioassay - urinalysis, NOT detected with a dosimeter (TLD or film).
9. ALI:
  - 10 mCi ( $4 \times 10^8$  Bq) inorganic compounds (vapor inhalation).
  - 2 mCi ( $7 \times 10^7$  Bq) by inhalation, weekly clearance.
  - 6 mCi ( $2 \times 10^8$  Bq) by ingestion.
10. Shielding: 3mm of plexiglass (if needed).

**Instructor's Notes**

11. Special Considerations:

- Detection of S-35 by radiation survey instruments requires special care due to the low efficiency of detection.
- Sulfur-35 compounds, including methionin, generate volatile fractions particularly during lyophilization or incubation.



**CHLORINE-36 (Cl-36)**

Chlorine-36 is a medium-energy beta emitter. Use a thin-end window G-M detector or liquid scintillation counter for detection.

1. Maximum energy: 0.710 MeV (the average energy is 0.233 MeV).
2. Maximum range in air: 7 feet (2 m).
3. Maximum range in tissue: 0.1 inch (2.6 mm).
4. IAEA toxicity classification: High-medium.
5. Physical half-life:  $3 \times 10^5$  years.
6. Effective half-life ( $T_{\text{eff}}$ ): 10-29 days.
7. Critical organ:
  - Whole body for transportable compounds.
  - Lung for inhalation.
  - Lower large intestine for ingestion.
8. Personnel monitoring: Urinalysis, finger rings.

**Instructor's Notes**

9. ALI:
- 0.2 mCi ( $7 \times 10^6$  Bq) by inhalation.
  - 2 mCi ( $7 \times 10^7$  Bq) by ingestion.
10. Shielding: 0.25 inches (6mm) of plexiglass.
11. Special Considerations:
- Cl-36 beta particles have sufficient energy to penetrate gloves and skin.
  - When handling millicurie quantities, do not work over an open container.
  - Avoid glove and skin contamination or ensure that it is promptly detected and removed.

**CALCIUM-45 (Ca-45)**

Calcium-45 is a low-energy beta emitter and may be detected with a thin-window probe. Calcium-45 is commonly used with animal studies.

1. Maximum beta energies: 0.257 MeV (the average energy is 0.086 MeV).
2. Maximum range in air: 20 inches (52 cm).
3. Maximum range in tissue: 0.62mm.
4. IAEA toxicity classification: High.
5. Physical half-life: 163 days.
6. Effective half-life ( $T_{\text{eff}}$ ): 163 days.
7. Critical organ: Bone.
8. Personnel monitoring: Bioassay, initially by urine, later by feces: NOT detected with a dosimeter (TLD or film).
9. ALI:
  - 0.8 mCi ( $3 \times 10^7$  Bq) by inhalation.
  - 2 mCi ( $7 \times 10^7$  Bq) by ingestion.
10. Shielding: 3mm of plexiglass.

**Instructor's Notes**

11. Special Considerations:

Detection of Ca-45 by radiation survey instruments requires special care due to the low efficiency of detection.

**CHROMIUM-51 (Cr-51)**

Chromium-51 is a gamma and a X-ray emitter. Cr-51 is readily detected with a thin-window G-M probe. Liquid scintillation counting is also used.

1. Maximum energy: 0.32 MeV gamma ray (9.8%), a very low energy (0.005 MeV) X-ray (22 %) and 0.004 MeV (66.9%) auger electron.
2. IAEA toxicity classification: Medium-low.
3. Physical half-life: 27.7 days.
4. Effective biological half-life: 27 days.
5. Critical organ: Lower large intestine, and lungs.
6. Personnel monitoring: Dosimeter, internal uptakes may be determined by urine or fecal sampling.
7. ALI:
  - 20 mCi ( $7 \times 10^8$  Bq) by inhalation, yearly clearance.
  - 20 mCi ( $7 \times 10^8$  Bq) by ingestion.
8. Shielding: - 3.2 mm of lead is the half value layer (thickness of lead that will reduce the dose rate by one-half).
9. Special Considerations:  
Use thin-end window G-M or solid scintillation detectors or liquid scintillation counting.

**IRON-55 (Fe-55)**

Iron-55 decays by electron capture and so is an X-ray emitter. Fe-55 contamination may be detected by a thin-window G-M probe looking at the very low-energy characteristic Mn X-rays. Manganese is formed when the iron nucleus captures an electron. The manganese emits X-rays characteristic to its electron shell structure. Liquid scintillation counting may also be used.

1. Average X-ray energy: Electron capture with an average low energy of 0.006 MeV.
2. IAEA toxicity classification: Medium-low.
3. Physical half-life: 2.6 years.
4. Effective half-life ( $T_{\text{eff}}$ ): 370 days.
5. Critical organ:
  - Liver and spleen for inhalation.
  - Lower large intestine for ingestion.
6. Personnel monitoring: Uptakes evaluated by analysis of blood.
7. ALI:
  - 2 mCi ( $7 \times 10^7$  Bq) by inhalation, daily Clearance.
  - 9 mCi ( $3 \times 10^8$  Bq) by ingestion.

**COBALT-57 (Co-57)**

Cobalt-57 is an X-ray emitter. Most common means of detection is with a thin-window G-M probe.

1. Maximum energy: X-ray radiation from 0.014 to 0.692 MeV (0.122 MeV emitted 85.5% of the time).
2. IAEA toxicity classification: Medium-low.
3. Physical half-life: 270.9 days.
4. Effective half-life ( $T_{\text{eff}}$ ): 9 days.
5. Critical organ: Lower large intestine.
6. Personnel monitoring: Dosimeter, uptakes may be evaluated by whole body counting.
7. ALI:
  - 0.7 mCi ( $3 \times 10^7$  Bq) by inhalation, yearly clearance.
  - 4 mCi ( $2 \times 10^8$  Bq) by ingestion.
8. Shielding: 3.2 mm of lead is the half value layer.

**IRON-59 (Fe-59)**

Iron-59 is a beta and gamma emitter that can create an external, an internal, and skin and eye hazard. Iron-59 is detected with a thin- end window G-M probe, solid scintillator, or liquid scintillation counter.

1. Maximum beta energies:
  - 0.466 MeV, average energy is 0.155 MeV.
  - 0.273 MeV, average energy is 0.091 MeV.
  - 0.131 MeV, average energy is 0.044 MeV.
2. Gamma energies:
  - 1.292 MeV.
  - 1.099 MeV.
  - 0.192 MeV.
  - 0.143 MeV.
3. Maximum range in air of beta: 45 inches (115 cm).
4. Unshielded dose rate from 1 mCi point source at ½ inch (1 cm): 6.18 rad/hr.
5. IAEA toxicity classification: Medium-high.
6. Physical half-life: 44.6 days.
7. Effective half-life ( $T_{\text{eff}}$ ): 42 days.



8. Critical organ:
  - Liver and spleen for inhalation.
  - Lower large intestine for ingestion.
9. Personnel monitoring: Dosimeter, finger rings - fecal analysis may be used to determine uptake for weeks or months after handling. Urinalysis is recommended from 4-24 hours after handling.
10. ALI:
  - 0.3 mCi ( $1 \times 10^7$  Bq) by inhalation.
  - 0.8 mCi ( $3 \times 10^7$  Bq) by ingestion.
11. Shielding: 0.38 inch (9.7 mm) of lead is the half value layer.
12. Special considerations:
  - Near an unshielded Fe-59 source, dose rates from beta radiation can be much higher than dose rates due to gamma radiation.
  - Avoid direct eye exposure.
  - Avoid skin exposure.

**IODINE-125 (I-125)**

Iodine-125 is a gamma and X-ray emitter. I-125 contamination may be detected by a thin-window G-M probe or with liquid scintillation counting.

1. Maximum energy: 0.035 MeV gamma (6.5%), 0.027 MeV x-ray (112.5%) and 0.031 MeV x-ray (25.4%).
2. IAEA toxicity classification: Medium-high.
3. Physical half-life: 60 days.
4. Effective half-life ( $T_{\text{eff}}$ ): 42 days.
5. Critical organ: Thyroid gland.
6. Personnel monitoring: Internal uptakes evaluated by thyroid scan.
7. ALI:
  - 0.06 mCi ( $2 \times 10^6$  Bq) by inhalation, daily clearance.
  - 0.04 mCi ( $1 \times 10^6$  Bq) by ingestion.
8. Shielding: 0.25 mm of lead is the half-value layer.
9. Other considerations for iodine compounds:
  - Volatilization of iodine (NaI) is the most significant hazard.

**Instructor's Notes**

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- Simply opening a vial of sodium iodide at high-radioactive concentrations can cause minute droplets to become airborne.
- Solutions containing iodide ions should not be made acidic nor stored frozen; both lead to formation of volatile elemental iodine.
- Some iodide compounds can penetrate surgical rubber gloves - wear two pairs or polyethylene gloves over rubber.
- Can be easily absorbed through the skin.

**Instructor's Notes****IODINE-131 (I-131)**

Iodine-131 is a gamma, X-ray, and beta emitter. I-131 contamination may be detected by a thin-window G-M probe or with liquid scintillation counting.

1. Maximum beta energies: 0.248 - 0.606 MeV.
2. Primary gamma energies: 0.364 MeV, 0.637 MeV, and 0.284 MeV.
3. IAEA toxicity classification: Medium-high.
4. Physical half-life: 8 days.
5. Effective half-life ( $T_{\text{eff}}$ ): 7.6 days.
6. Critical organ: Thyroid gland.
7. Personnel monitoring: Dosimeter, thyroid scan for uptakes.
8. ALI:
  - 0.05 mCi ( $2 \times 10^6$  Bq) by inhalation, daily clearance.
  - 0.03 mCi ( $1 \times 10^6$  Bq) by ingestion.
9. Shielding: 2.3 mm of lead is the half-value layer.

Note: gammas are by decreasing percentages.

10. Other considerations for iodine compounds:

Volatilization of iodine is the most significant hazard.

- Simply opening a vial of sodium iodide (NaI) at high-radioactive concentrations can cause minute droplets to become airborne.
- Solutions containing iodide ions should not be made acidic nor stored frozen; both lead to formation of volatile elemental iodine.
- Some iodide compounds can penetrate surgical rubber gloves - wear two pairs or polyethylene gloves over rubber.
- Can be easily absorbed through the skin.

## APPENDIX B

## STORAGE OF RADIOACTIVE/HAZARDOUS MATERIALS

**1.0 Containment of Material**

Containment generally means using vessels, trays, diaper paper, bench tops, etc. to contain contamination.

**2.0 Segregation and Storage**

Segregate incompatibles and store by hazard class.

Recommended general hazard classes for storage are:

1. Caustics (bases).
2. Acids (mineral).
3. Flammables (including organic acids).
4. Poisons (toxics).
5. Oxidizers.
6. Water reactives.

**3.0 General Guidelines**

1. Keep flammables by themselves in Underwriters Laboratory (UL) or Factory Mutual (FM) approved safety cans or cabinets.
2. Keep acids away from bases.
3. Separate organics from inorganics.
4. Store oxidizers away from flammables.

**Instructor's Notes**

5. Provide as much physical separation as possible between classes.
6. Biohazards should be properly labeled and may be stored as one group.
7. Class B and C carcinogens should be properly labeled and stored with their chemical family.
8. Store Class A carcinogens in the glovebox or another regulated area.

**Instructor's Notes**

Note: EPA carcinogen designations are as follows:  
EPA-A Human Carcinogen: sufficient evidence from epidemiologic studies to support a casual association between exposure and cancer.  
EPA-B Probable Human Carcinogen: weight of evidence of human carcinogenicity based on epidemiologic studies is limited; agents for which weight of evidence of carcinogenicity based on animal studies is sufficient.  
Two subgroups: B1: Limited evidence of carcinogenicity from epidemiologic studies;  
B2: Sufficient evidence from animal studies; inadequate evidence or no data from epidemiologic studies.

**Instructor's Notes**

EPA-C: Possible Human Carcinogen: Limited evidence of carcinogenicity in animals in the absence of human data.

**4.0 Formation of Organic Peroxides**

Organic peroxides are a class of compounds that have unusual stability problems that make them among the most hazardous substances found in the laboratory. As a class, organic peroxides are considered to be powerful explosives and are sensitive to heat, friction, impact, and light, as well as to strong oxidizing and reducing agents. Common compounds that form peroxides during storage include:

- ethyl ether.
- isopropyl ether.
- potassium metal.
- vinyl chloride.
- cyclohexene.
- dicyclopentadiene.
- vinyl acetylene.
- dioxane.
- acetal.
- butadiene.
- vinyl ethers.
- styrene.
- diacetylene.
- vinyl acetate.



- tetrahydrofuran.
- divinylidene chloride.
- cumene.
- sodium amide.
- methyl acetylene.
- methylcyclopentene.

**Instructor's Notes**

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**APPENDIX C**  
**STORAGE GUIDELINES BY HAZARD CLASS**

**Instructor's Notes**

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**1.0 Caustics**

Caustics are materials with  $\text{pH} > 10$ . Examples include ammonium hydroxide, calcium hydroxide, and sodium hydroxide.

- 1.1 Separate from acids.
- 1.2 Store solutions of inorganic hydroxides in polyethylene containers.
- 1.3 Store large containers below eye level.

**2.0 Acids**

Acids are materials with  $\text{pH} < 5$ . Examples include acetic, chromic, and hydrofluoric.

- 2.1 Separate from bases and materials that could evolve toxic vapors on contact (i.e., sodium cyanide).
- 2.2 Store large bottles low to the ground -- at least below eye level.
- 2.3 Separate inorganic acids from organic acids (i.e., acetic, oxalic, etc.).
- 2.4 Separate from active metals (i.e., sodium, potassium).
- 2.5 Store perchloric and nitric acid as oxidizers.

**3.0 Flammables/Combustibles**

Flammables/combustibles vapors ignite easily at room temperature. Examples include alcohols, esters, ketones, ethers, and pyrophorics.

- 3.1 Store flammable liquids in Underwriters Listed (UL) or Factory Mutual (M) safety cans or cabinets.
- 3.2 Keep away from heat, sun, flame, and spark sources.
- 3.3 Separate from oxidizers.
- 3.4 Use only (UL) listed or FM approved “explosion safe” or “spark-proof” refrigerators for cold storage of flammables.

**4.0 Poisons (Toxics)**

Poisons are dangerous if inhaled, swallowed, or absorbed through the skin. Examples include phenol and hydrazine.

- 4.1 Store according to label directions.
- 4.2 Separate from other hazard classes.
- 4.3 Keep tightly sealed.

**5.0 Oxidizers**

Oxidizers are materials that yield oxygen: react with water, fire, flammables, and combustibles. Examples include inorganic nitrates, permanganates, inorganic peroxides, persulfates, and perchlorates.

Oxidizers must be stored in accordance with NFPA 45, *Standard on Fire Protection for Laboratories Using Chemicals* and NFPA 430, *Code for the Storage of Liquid and Solid Oxidizers*.

**Instructor's Notes**

- 5.1 Keep separate from flammables and other organic materials.
- 5.2 Keep separate from reducing agents (i.e., zinc, alkaline metals, formic acid).
- 5.3 Do not store directly on wooden surfaces.
- 5.4 Peroxide formers should be labeled with date received and opened, and should be discarded as hazardous waste within three to six months of opening. Depending on the chemical, unopened peroxide performers should be discarded within 12 months of receipt.

**Instructor's Notes**

**6.0 Organic Peroxides**

Organic peroxides are a class of compounds that have unusual stability problems.

Oxidizers must be stored in accordance with NFPA 45, *Standard on Fire Protection for Laboratories Using Chemicals* and NFPA 432, *Code for the Storage of Organic Peroxides*.

## APPENDIX D

Lesson Plan	Instructor's Notes
<p><b>I. RELEASE OF MATERIALS</b></p> <p>A. Monitoring Techniques General</p> <p>Monitoring techniques for release of materials are addressed in DOE/CH-9401 (1993). The following techniques apply for G-M detectors (H-3 cannot be measured).</p> <ol style="list-style-type: none"> <li>1. Surveys should be conducted in a low background area (background levels are not to exceed 300 cpm; lower levels are preferable and in most cases achievable).</li> <li>2. Direct measurement should be made prior to smear surveys.</li> <li>3. Materials or equipment with inaccessible surface areas should be disassembled for survey or the inaccessible areas evaluated for contamination with special survey techniques or by review of process knowledge.</li> </ol> <p>If potential for internal contamination cannot be adequately assessed, material may not be released.</p> <ol style="list-style-type: none"> <li>4. An audible response should be utilized as the principal indicator for initial detection of surface radioactivity.</li> </ol>	

Lesson Plan	Instructor's Notes
<p>5. The assigned instrument/detector efficiencies should reflect a prior evaluation of facility wastes.</p> <p>Typical efficiencies for a thin-window G-M probe.</p> <ul style="list-style-type: none"> <li>• C-14, S-35 - 10%.</li> <li>• P-32 - 50%.</li> </ul> <p>B. Beta/Gamma Direct Monitoring</p> <ol style="list-style-type: none"> <li>1. <u>Window</u>: Use a thin-window probe, detector window thickness (mylar) should not be more than 2.0 mg/cm<sup>2</sup>.</li> <li>2. <u>Scanning</u>: Scan the surface; in most cases, scanning will cover nearly 100 percent of accessible surfaces.</li> <li>3. <u>Distance</u>: Maintain detector window no more than inch from surface.</li> <li>4. <u>Speed</u>: The number of counts produced in the detector is inversely proportional to the scanning speed.</li> <li>5. <u>Audio</u>: If at any point a perceivable audible or visual response is detected, perform a stationary evaluation of count rate.</li> </ol>	<p>Results: If surveys indicate presence of contamination (refer to 10 CFR 835 Appendix D for release from controlled areas or DOE 5400.5, Ch.2, for unrestricted release), the material should not be released. Prior to release, contamination levels must be less than appropriate values and should be reduced as low as reasonably achievable.</p>

Lesson Plan	Instructor's Notes
<p data-bbox="261 262 781 300">C. Smear Surveys for Releasing Material</p> <ol data-bbox="334 331 997 667" style="list-style-type: none"><li data-bbox="334 331 997 401">1. An initial screening evaluation may be conducted by wiping 100 percent of surface.</li><li data-bbox="334 432 997 501">2. These large-area wipes may be evaluated by holding the probe up to the swipe (~5 sec.).</li><li data-bbox="334 533 997 667">3. If initial screening evaluation indicates presence of contamination, take representative disc smears (100 cm<sup>2</sup> spill area) of up to 100 percent of accessible surface areas.</li></ol> <p data-bbox="261 699 518 737">D. Documentation</p> <p data-bbox="334 768 979 869">All surveys for release shall be documented in writing. Documentation of release from controlled areas should include information required by RCS Article 421.5.</p>	



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## APPENDIX E

## IDENTIFICATION OF MIXED LOW-LEVEL RADIOACTIVE WASTE (LLW)

	Instructor's Notes
<p><b>1.0 40 CFR PART 261</b></p> <p>RCRA, in 40 CFR Part 261 Subpart C, defines general characteristics that, if exhibited by a waste material, require the classification of that material as hazardous. These characteristics are:</p> <ul style="list-style-type: none"><li>1.1 Ignitability.</li><li>1.2 Corrosivity.</li><li>1.3 Reactivity.</li><li>1.4 Toxicity.</li></ul>	
<p><b>2.0 Specific Waste Streams</b></p> <p>In addition to defining the characteristics of hazardous wastes, 40 CFR Part 261 lists specific waste streams that are considered hazardous. These lists are compiled in tables in Subpart D according to:</p> <ul style="list-style-type: none"><li>2.1 Hazardous waste from nonspecific.</li><li>2.2 Hazardous waste from specific sources.</li><li>2.3 Discarded commercial chemical products, off specification species, and container and spill residues.</li></ul>	

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**APPENDIX F**  
**RELEASE OF POTENTIALLY CONTAMINATED MATERIAL – DOE ORDER**  
**5400.5, Ch. 2**

**Instructor's Notes**

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**1.0 Surface Contamination Levels**

Prior to being released, property should be surveyed to determine whether both removable and total surface contamination (including contamination present on and under any coating) is greater than the levels given in the DOE Order 5400.5, Ch. 2.

**2.0 Potential For Contamination**

Property should be considered to be potentially contaminated if it has been used or stored in radiological areas that could contain unconfined radioactive material or that are exposed to beams of particles capable of causing activation.

Material and Equipment in Radiological Areas established to control surface or airborne radioactive material shall be treated as potentially contaminated.

**3.0 Inaccessible Areas**

Where potentially contaminated surfaces are not accessible for measurement, such property may be released after case by case evaluation and documentation based on both the history of its use and available measurements demonstrate that the unsurveyable surfaces are likely to be within the release criteria.

**4.0 Volume Contamination**

EH-412 has provided guidance for release of material that has been contaminated with tritium (reference DOE 1995). Other materials may be released if criteria and survey techniques are approved by DOE.

**5.0 Items With Fixed Contamination**

Under exceptional conditions, materials and equipment with fixed contamination that exceeds the release criteria may be released for use in Controlled Areas outside of Radiological Areas. As a condition of such release, the removable contamination levels must be below the level specified in Appendix D of 10 CFR 835. The materials shall be routinely monitored, clearly labeled and/or tagged to alert personnel of the contaminated status, and have appropriate administrative procedures established and exercised to maintain control of these items.

**6.0 Covering of Contaminated Surfaces**

Radioactivity on equipment should not be covered by paint, plating, or other covering material unless contamination levels, as determined by a survey and documented, are below the "Removable" Criteria of Table 1 of DOE Order 5400.5, Ch. 2 (see below). A reasonable effort must be made to minimize the contamination prior to use of any covering.

**Instructor's Notes**

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If it is likely that contamination exists under a painted surface, it may be necessary to remove some of the painted surface to measure contamination levels below. Use a paint remover to collect paint samples from areas of approximately 200 cm<sup>2</sup>. Measure alpha and/or beta-gamma levels beneath paint. Check with Radiological Control personnel prior to using any paint remover to eliminate the generation of mixed waste.

Where potentially contaminated surfaces are not accessible for measurement, the equipment may be released after case by case evaluation and documentation based on both the history of its use and available measurements demonstrate that the unsurveyable surfaces are likely to be within the release criteria.

**Allowable Total Residual Surface Contamination**  
**(dpm/100 cm<sup>2</sup>)<sup>1/</sup>**

Radionuclides <sup>2/</sup>	Average <sup>3/, 4/</sup>	Maximum <sup>4/, 5/</sup>	Removable <sup>4/, 6/</sup>
Transuranics, I-125, I-129, Ra-226,	500	1500	20
Ac-227, Ra-228, Th-228, Th-230, Pa-231, Th-Natural, Sr-90, I-126, I-131,	1,000	3,000	200
I-133, Ra-223, Ra-224, U-232, Th-232 U-Natural, U-235, U-238, and associated decay product, alpha emitters.	5,000	15,000	1,000
Beta-gamma emitters (radionuclides with decay modes other than alpha emission or spontaneous fission) except Sr-90 and others noted above. <sup>7/</sup>	5,000	15,000	1,000

**Figure IV-1 Surface Contamination Guidelines**

<sup>1/</sup> As used in this table, dpm (disintegrations per minute) means the rate of emission by radioactive material as determined by correcting the counts per minute measured by an appropriate detector for background, efficiency, and geometric factors associated with the instrumentation.

<sup>2/</sup> Where surface contamination by both alpha- and beta-gamma-emitting radionuclides exists, the limits established for alpha- and beta-gamma-emitting radionuclides should apply independently.

<sup>3/</sup> Measurements of average contamination should not be averaged over an area of more than 1 m<sup>2</sup>. For objects of less surface area, the average should be derived for each such object.

<sup>4/</sup> The average and maximum dose rates associated with surface contamination resulting from beta-gamma emitters should not exceed 0.2 mrad/h and 1.0 mrad/h, respectively, at 1 cm.

<sup>5/</sup> The maximum contamination level applies to an area of not more than 100 cm<sup>2</sup>.

<sup>6/</sup> The amount of removable material per 100 cm<sup>2</sup> of surface area should be determined by wiping an area of that size with dry filter or soft absorbent paper, applying moderate pressure, and measuring the amount of radioactive material on the wiping with an appropriate instrument of known efficiency. When removable contamination on objects of surface area less than 100 cm<sup>2</sup> is determined, the activity per unit area should be based on the actual area and the entire surface should be wiped. It is not necessary to use wiping techniques to measure removable contamination levels if direct scan surveys indicate that the total residual surface contamination levels are within the limits for removable contamination.

<sup>7/</sup> This category of radionuclides includes mixed fission products, including the Sr-90 which is present in them. It does not apply to Sr-90 which has been separated from the other fission products or mixtures where the Sr-90 has been enriched.

## APPENDIX G

### ANIMAL FACILITIES

#### Instructor's Notes

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#### 1.0 Objective

Describe the exposure potentials when handling animals that contain radionuclides during experiments.

#### 2.0 Outline

- Exposure potentials.
- Methods of minimizing external exposures.
- Methods of minimizing internal exposures.
- Methods of minimizing cross contamination.
- Room contamination problems.

#### 3.0 Dose Potentials

Animals that have had radionuclides administered to them can present a hazard both to the individuals working with them and to the success of the experiment as well. When using radionuclides in animals, one must be sure that the cages are well labeled, away from much foot traffic, and properly maintained to minimize the hazards arising from radionuclide use. Hazards that are likely to be encountered when using radionuclides in animals are listed below:

- 3.1 External personnel exposure resulting from gammaemitting radionuclides that have been used in an animal.
- 3.2 Internal radiation exposure resulting from accidental ingestion of radionuclides.
- 3.3 Cross contamination of radionuclides from one radiological experiment to another.



- 3.4 Room contamination that can result in the spread of radionuclides to a non-radioactive use area.

#### 4.0 Methods of Minimizing External Dose

##### 4.1 Distance

Maintain the greatest distance possible between the worker and the gamma-emitting animal to make full use of the inverse square law. This law, simply stated, implies that if the distance from a radioactive source is doubled, the dose is reduced by a factor of four.

##### 4.2 Time

When the distance cannot be minimized, the amount of time spent in the proximity of the radioactive animal should be kept at a minimum.

##### 4.3 Shielding

When the above is not possible for any reason or when the dose rate is determined to be very high, lead shielding of the proper thickness should be placed between the worker and animal. This is a clumsy technique and should only be attempted following consultation with a health physicist.

##### 4.4 Source Reduction

Following the experiment and/or the death of the animal, proper waste disposal proceedings should be initiated.

**5.0 Methods of Minimizing Internal Dose**

When handling radioactive animals or applying radionuclides to animals, it is required that the handler wear gloves that can be thrown away when contaminated. This will prevent the transfer of radionuclides from hand to mouth and is equally important when handling excreta or animal parts that may be radioactive.

Procedures for handling of the animal should be implemented to reduce the possibility of animal bites. This would include such activities as sedation of the animal in some procedures and special handling techniques for specific animals in other procedures. Both the researcher and the animal handler should be trained in these procedures.

**6.0 Methods of Minimizing Cross Contamination**

Cross contamination may cause the radioactive materials used in one experiment to turn up unexpectedly in the results of another. Since some experiments utilize only minute amounts of radionuclides while others use large amounts, it is easy to visualize the confusion when large amounts of unknown radionuclides appear suddenly in a low-level experiment.

**Instructor's Notes**

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Cross contamination can result from mishandling; contaminated protective clothing; contaminated cages, food, and water supplies; and airborne materials being transferred from cage to cage.

Contaminated cages should be allowed to decay where possible, and then be thoroughly washed. Following washing, they must be surveyed before re-use.

**7.0 Room Contamination Problems**

These problems can best be avoided by maintaining radioactive animals in a separate room and maintaining high standards of housekeeping in the room. Feces, cage linings, and urine should be stored in the appropriate containers. These items should not be allowed to accumulate.

In case any of the above-mentioned items are spilled, they should be cleaned up immediately, utilizing absorbent, disposable materials. All materials used in cleaning up a spill should be placed in the appropriate containers to preclude the possibility of further contamination spread. It is essential that spills be cleaned up without delay.

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